

ATOMIC-SCALE THEORY OF RADIATION-INDUCED PHENOMENA

OVERVIEW OF THE LAST FIVE YEARS
AND NEW RESULTS

Sokrates T. Pantelides

Department of Physics and Astronomy, Vanderbilt
University, Nashville, TN

The theory team:

Leonidas Tsetseris, Matt Beck, Ryan Hatcher, George Chatzisavvas,
Sasha Batyrev, Yevgenyi Puzyrev, Nikolai Sergueev, Blair Tuttle



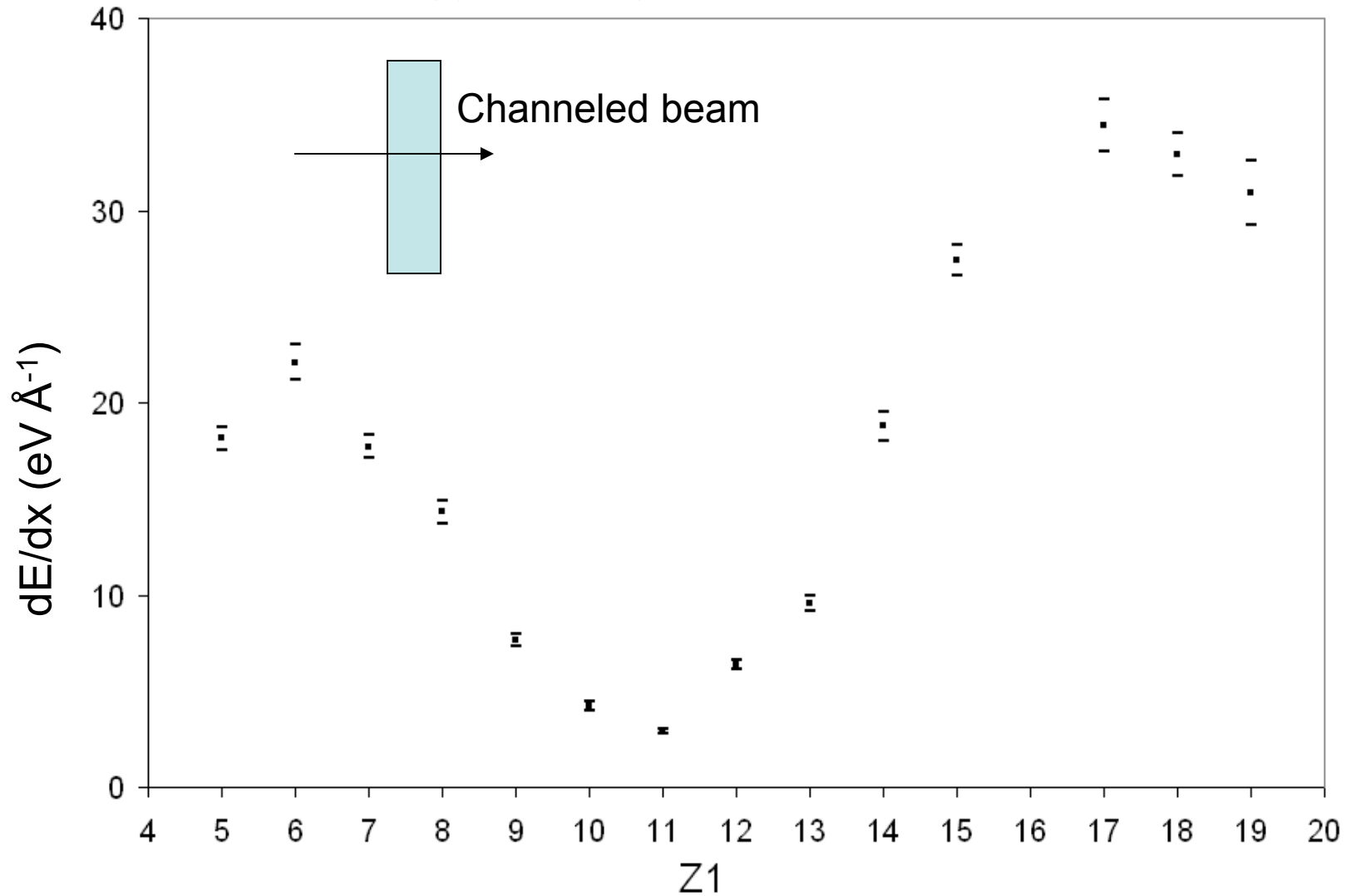
AFOSR/MURI FINAL REVIEW 2010



THEORY OBJECTIVES

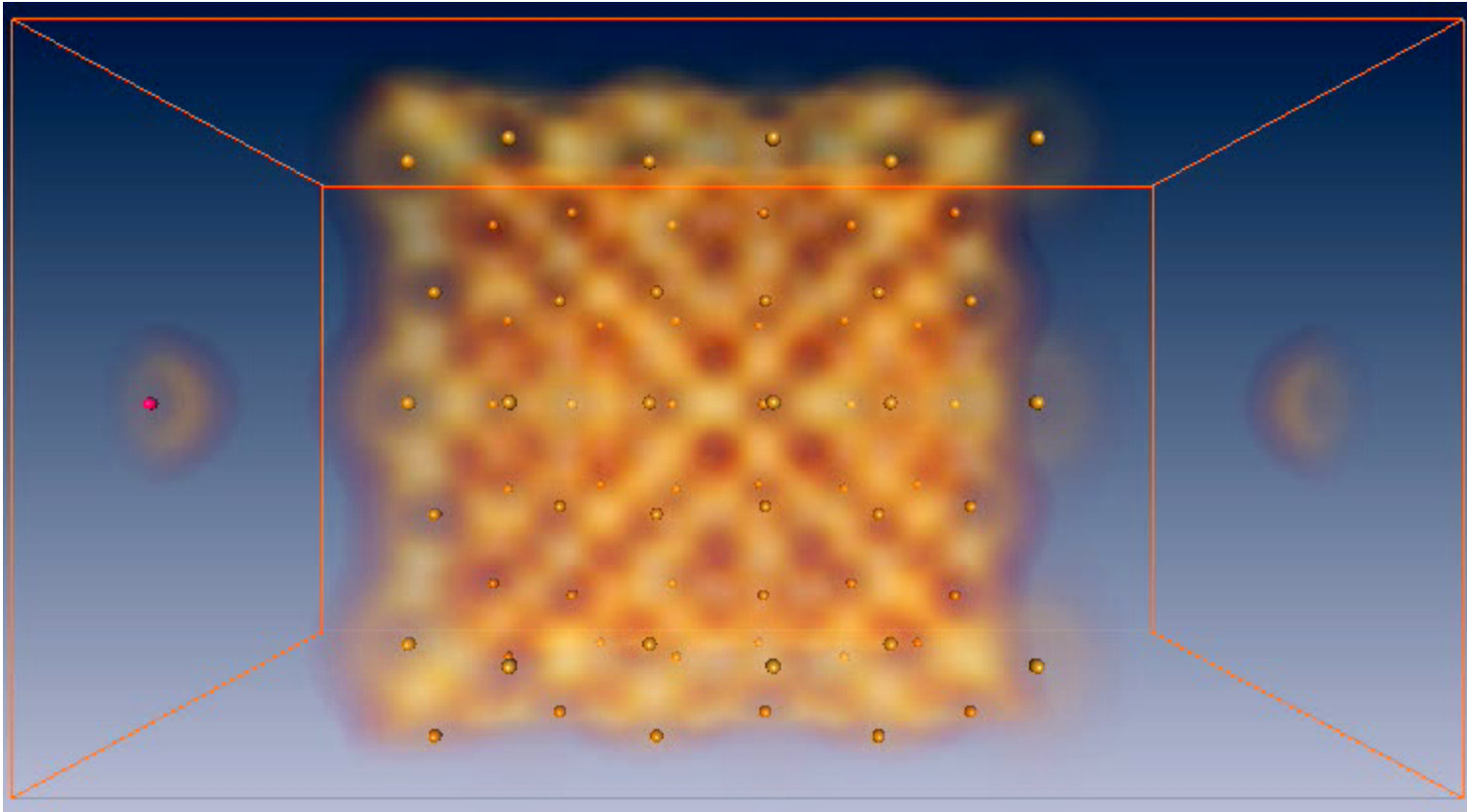
- DISPLACEMENT DAMAGE
 - Defects, charging
 - electrons
- ALTERNATE DIELECTRICS
 - Interface structure, interface defects, NBTI,...
- CARRIER MOBILITIES
- LEAKAGE CURRENTS

Energy loss by channeled ions in Si

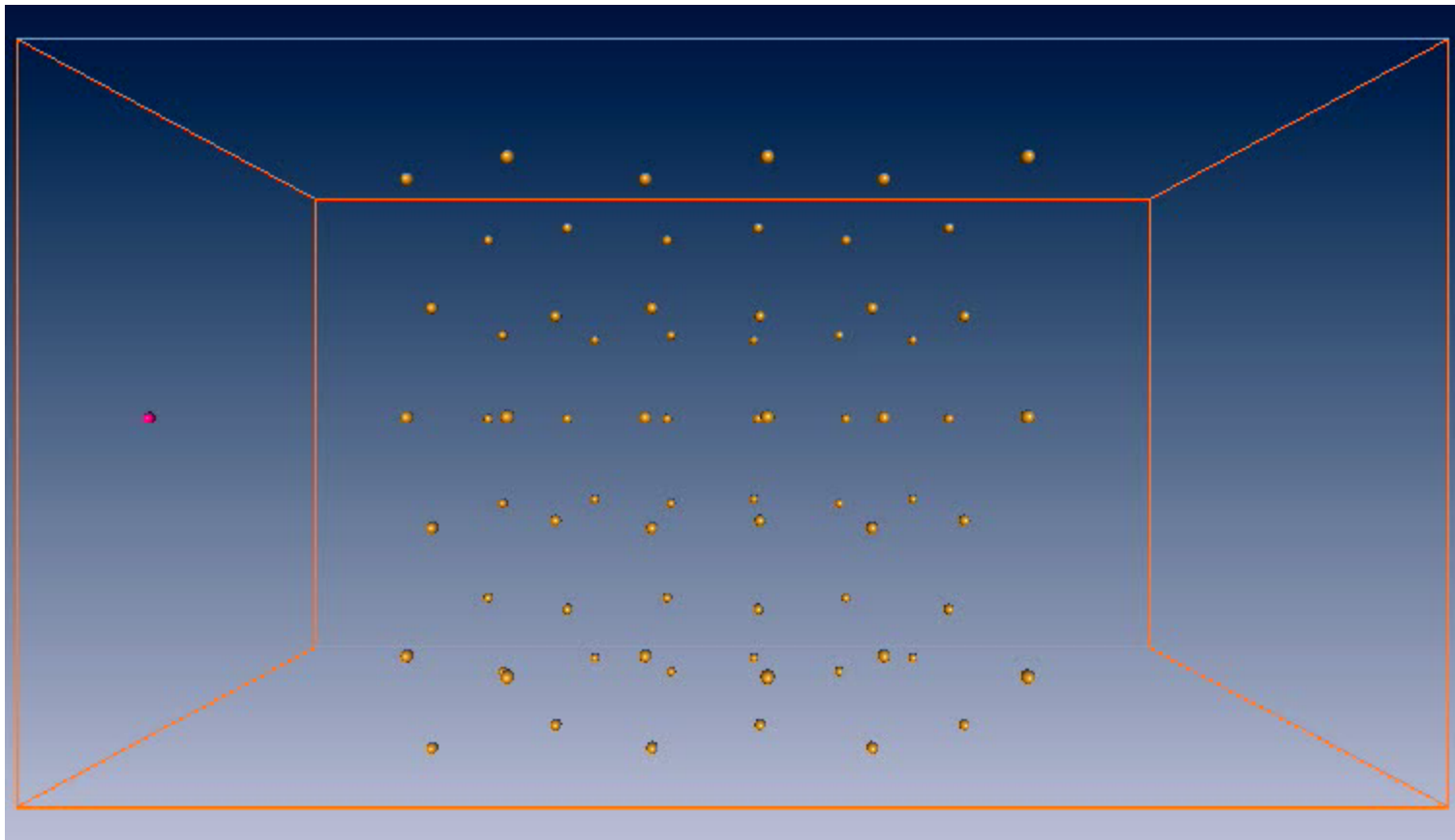


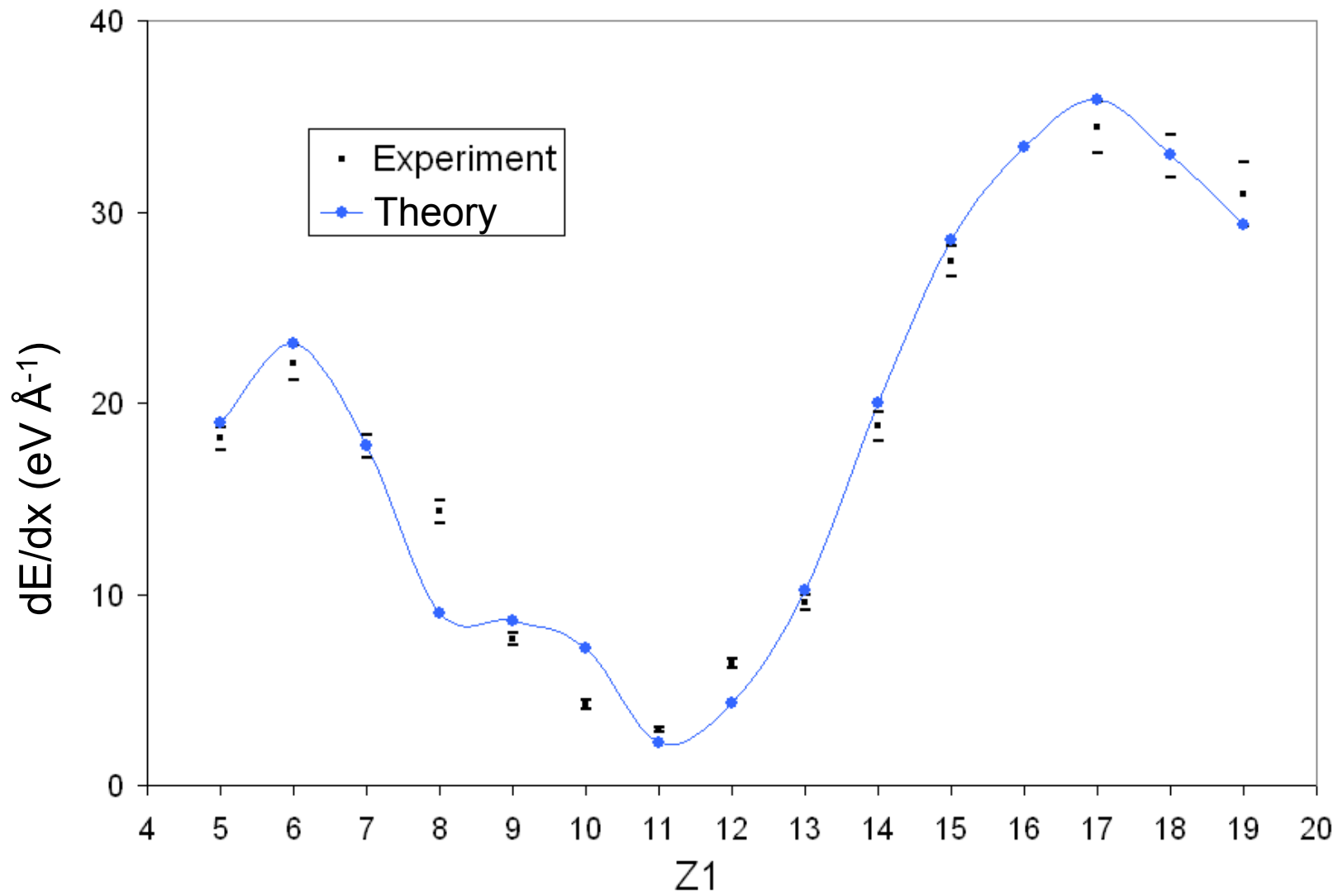
Experimental data, Eisen, 1968

Motion of an ion thru Si<110>



$$n(t) - n(0)$$

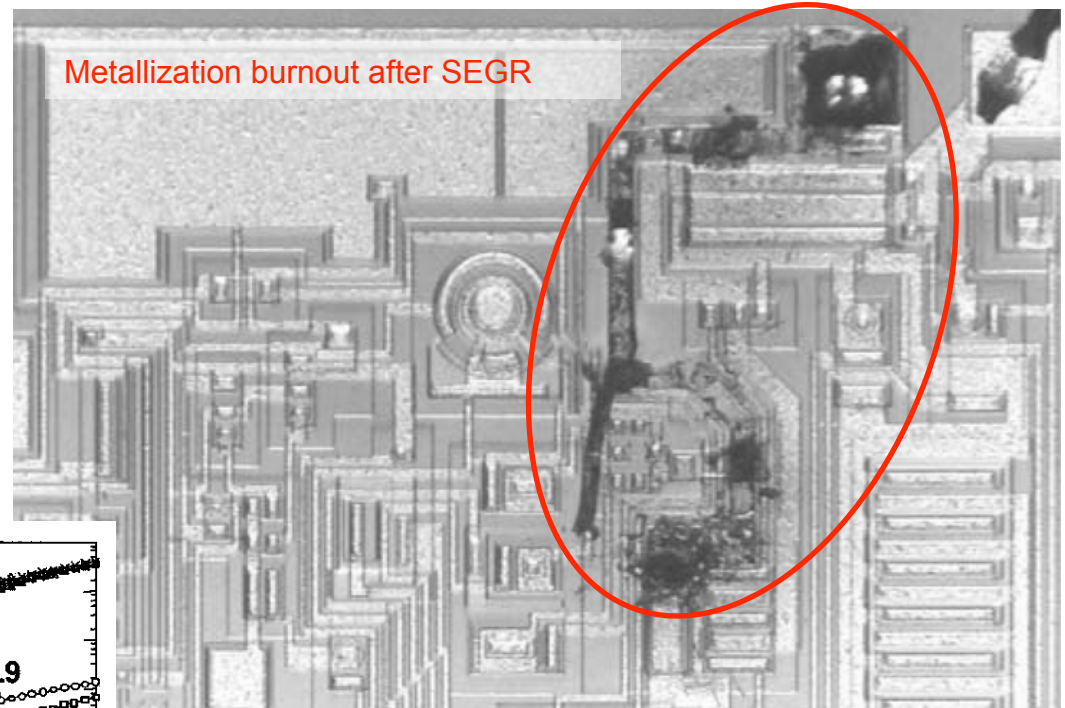




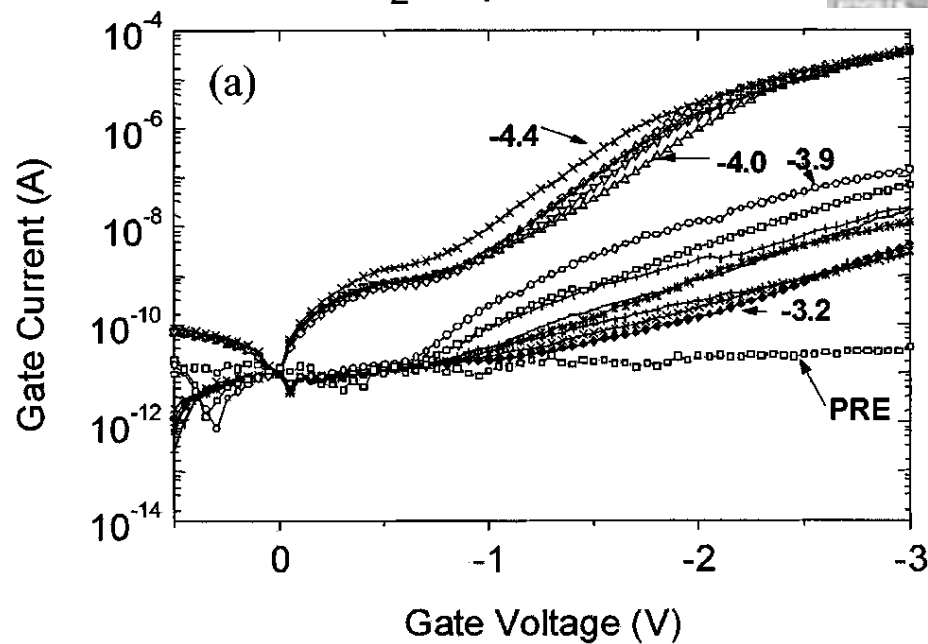
Data: Eisen (1968)

Ryan Hatcher

Single-Event Gate Rupture



I-V following biased irradiation of
3.3 nm SiO₂ capacitors



Lum et al., IEEE TNS (2004)

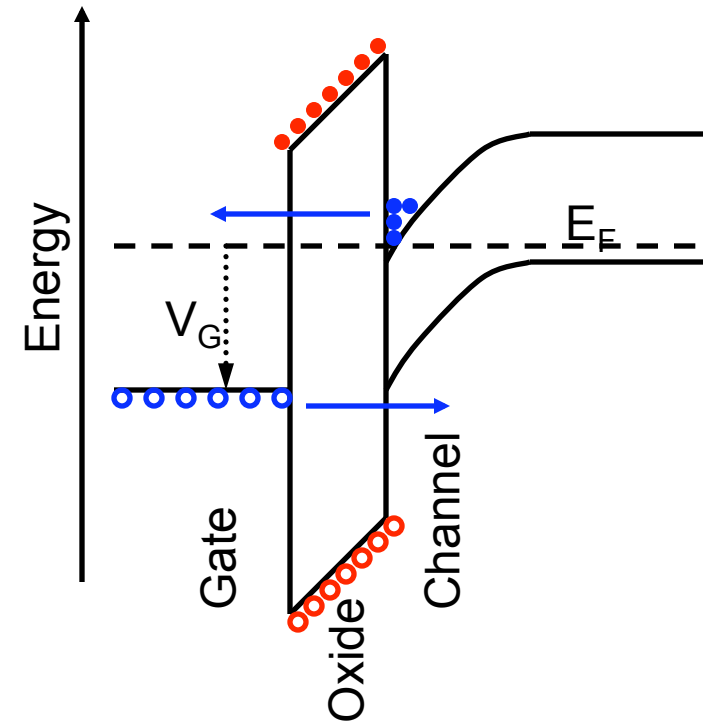
Massengill et al., IEEE TNS (2001)

Leakage-induced rupture

Breakdown results from local heating
due to ion-excited carriers...

...AND carriers injected by applied fields

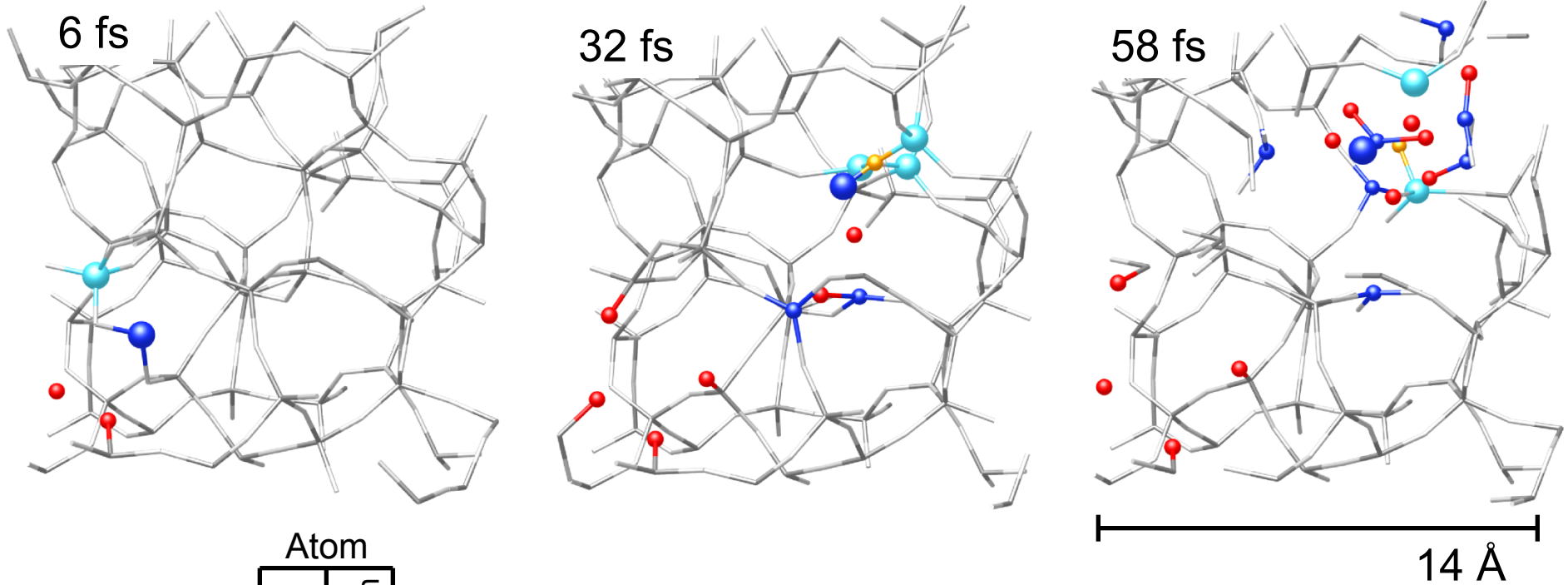
$$\frac{1}{E_{CR}} = \frac{q\mu_1 n(V)}{J_{CR}} + \frac{q\mu_2 n(L)}{J_{CR}}$$



Is it defect mediated?

Low-energy recoil dynamics in SiO₂

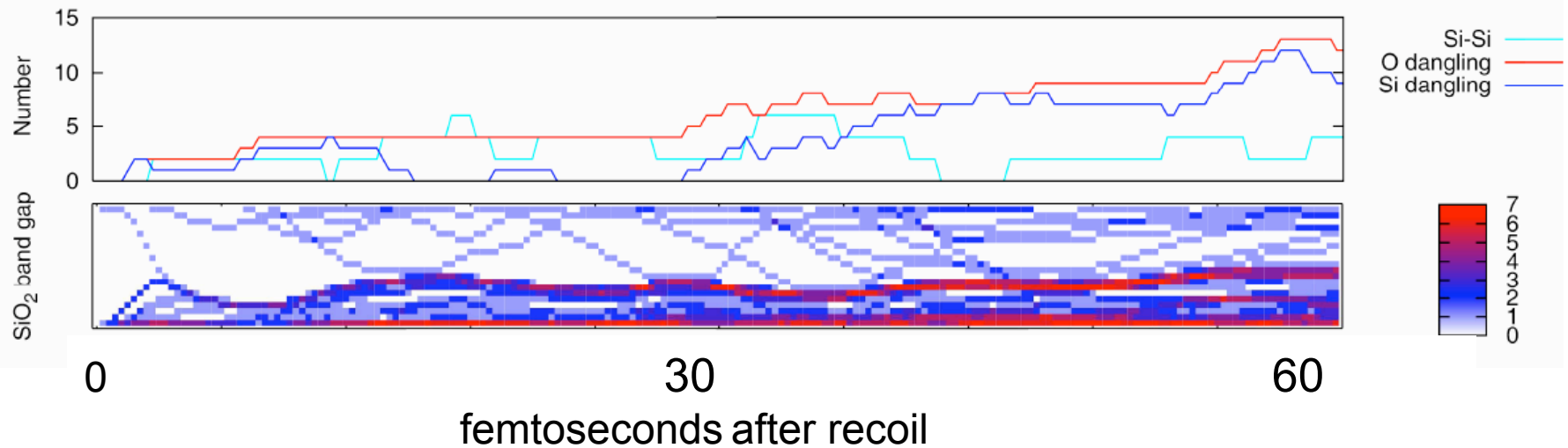
100 eV Si recoil



Defect	Atom	
	Silicon	Oxygen
Dangling Bond	●	●
Extra Bond	●	●
“Self Bond”	Big Ball	

Damage in amorphous material:
Network defects

Defect states in SiO₂

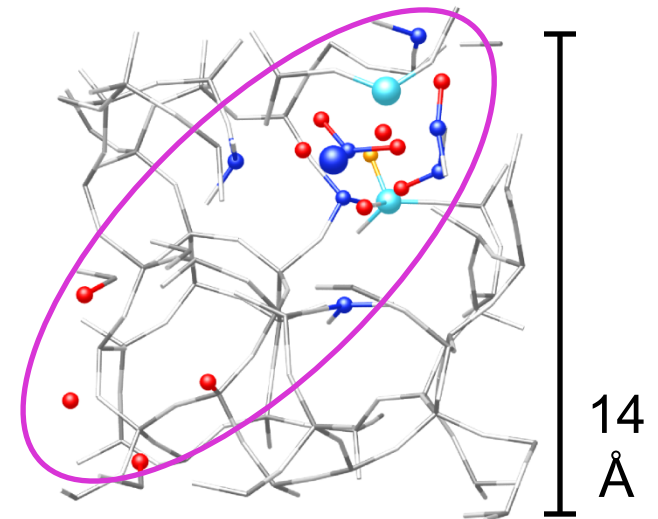


Increasing numbers of defects...

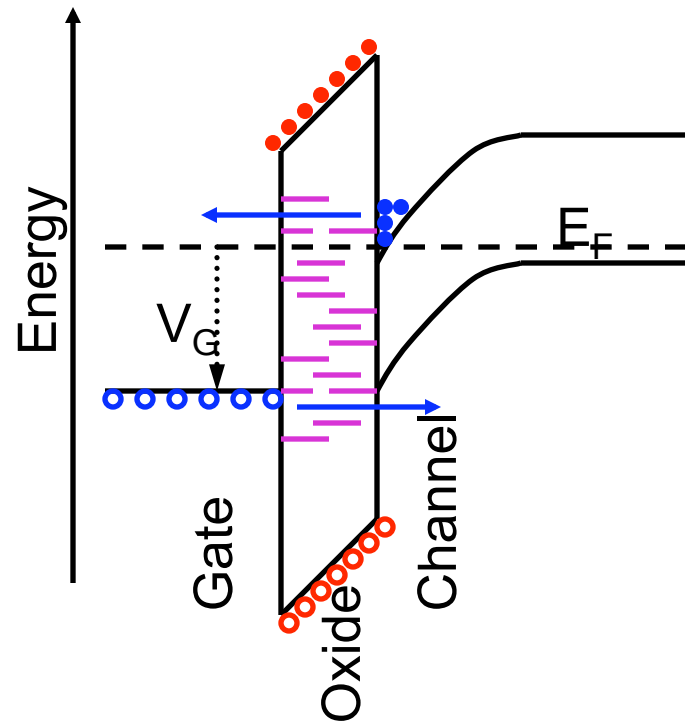
...increasing number of defect states within the bandgap!

Defect states separated by $\sim 2-5 \text{ \AA}$!

Conducting path!

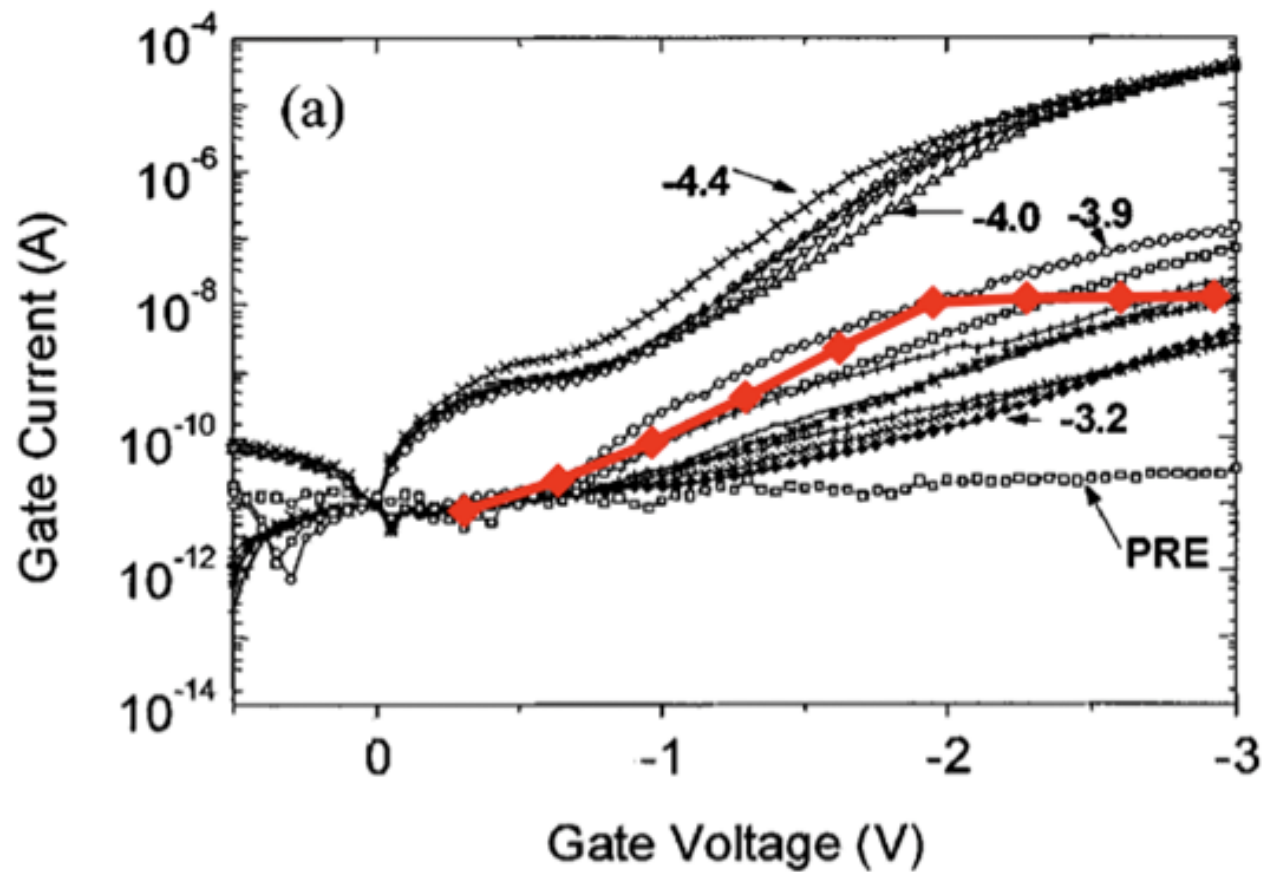


Defect-mediated leakage



Displacement-damage-induced defect states
facilitate field-injected leakage

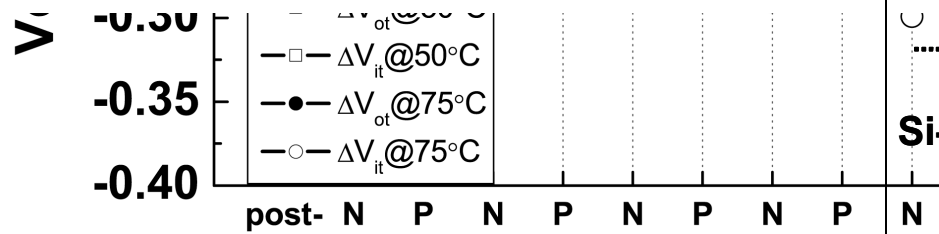
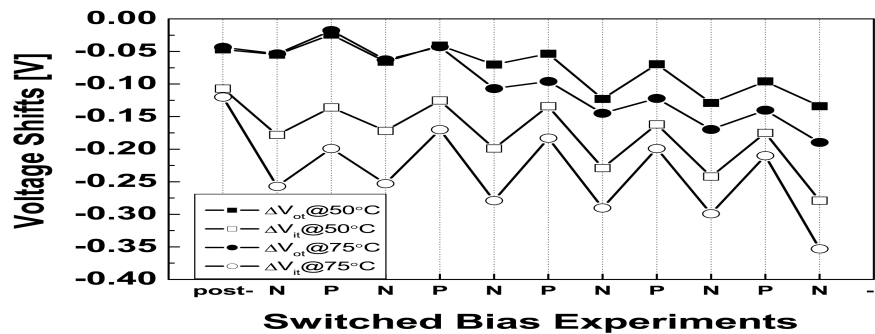
MULTISCALE TRANSPORT THEORY



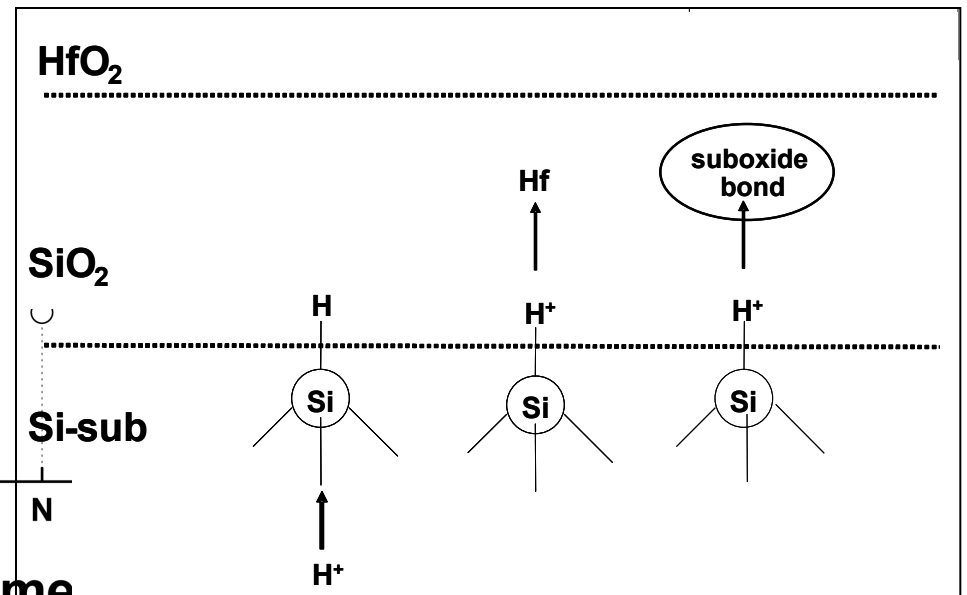
Data by Massengill et al. 2001

Si-SON-HfO₂

POST-IRRADIATION SWITCH-BIAS ANNEALING



Switched Bias Experiments

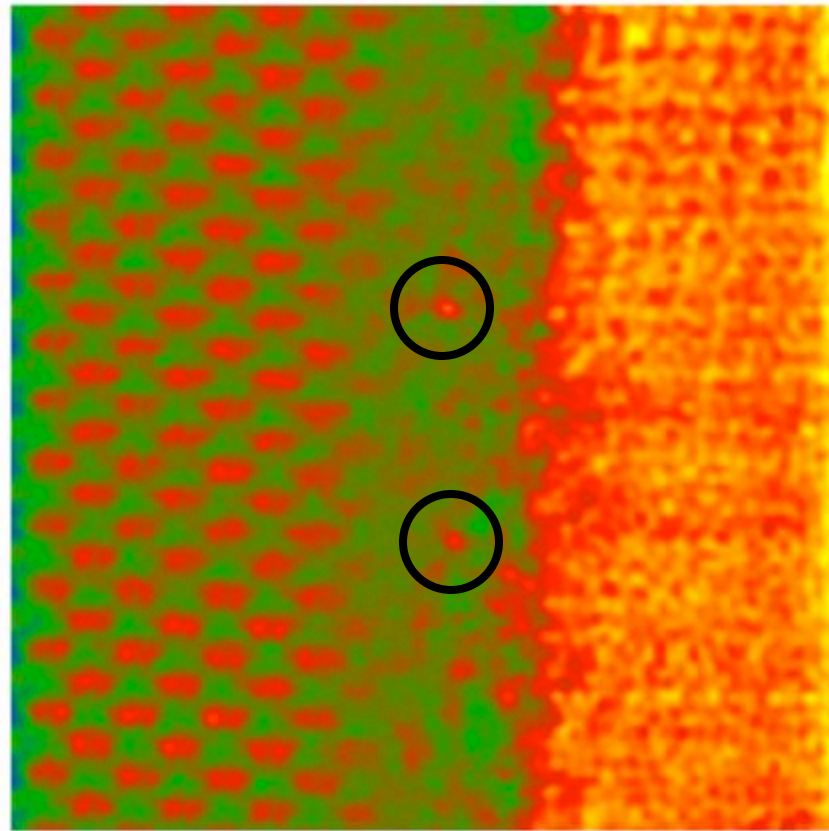


Vanderbilt team

Si

SiO₂

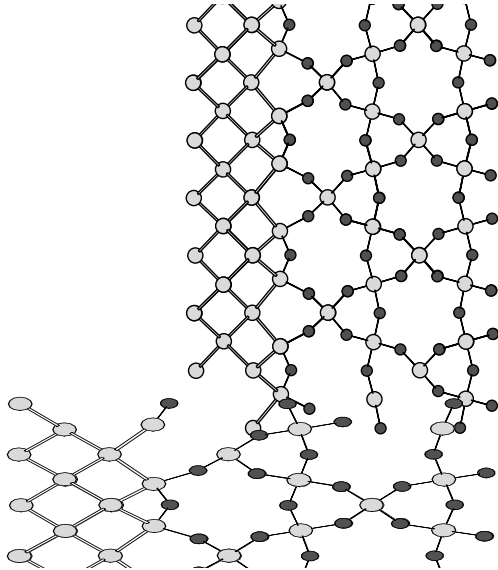
HfO₂



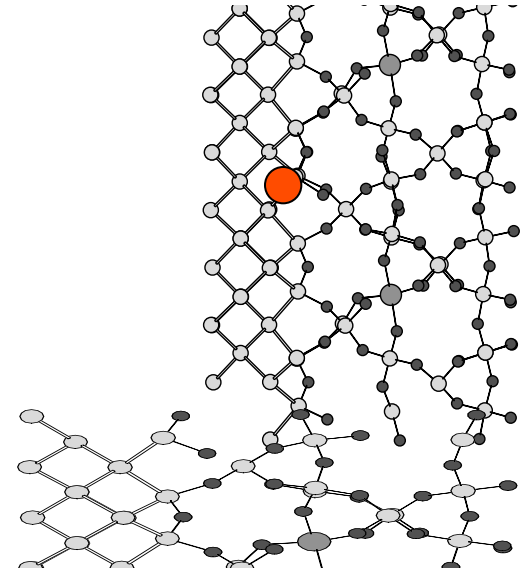
8 Å

Van Benthem & Pennycook, ORNL

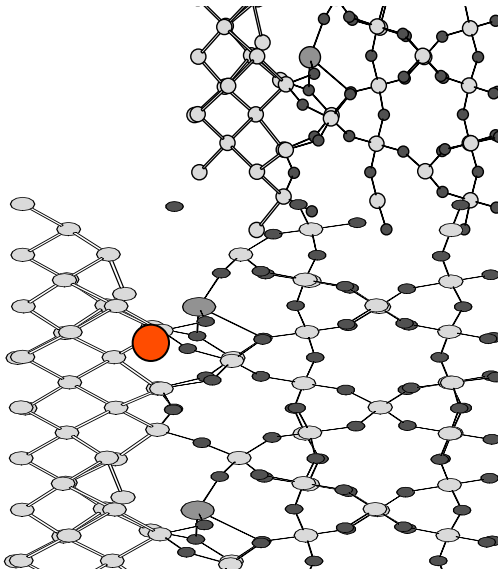
perfect structure



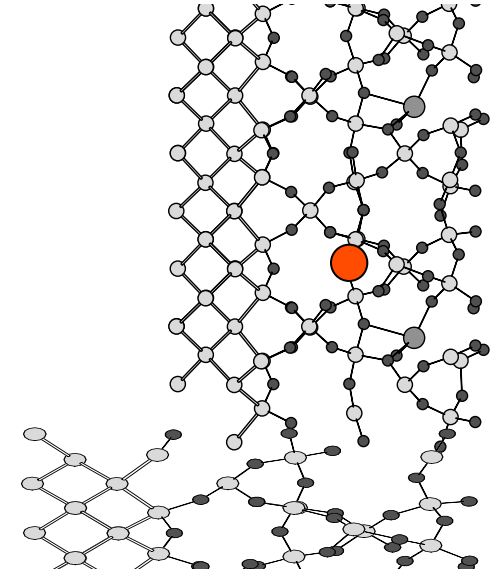
substitutional Hf: local lattice expansion



$$E(\text{int}) = E(\text{sub}) + 5 \text{ eV}$$



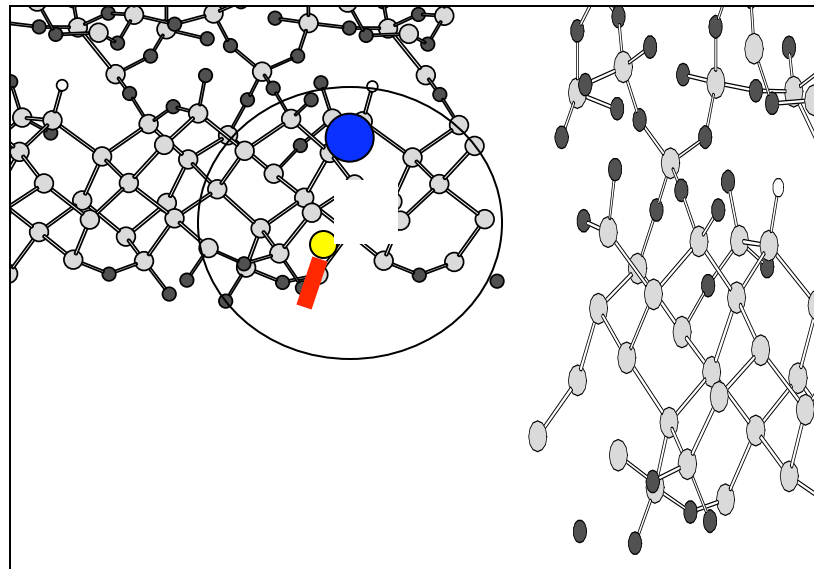
interstitial Hf:
rebonding



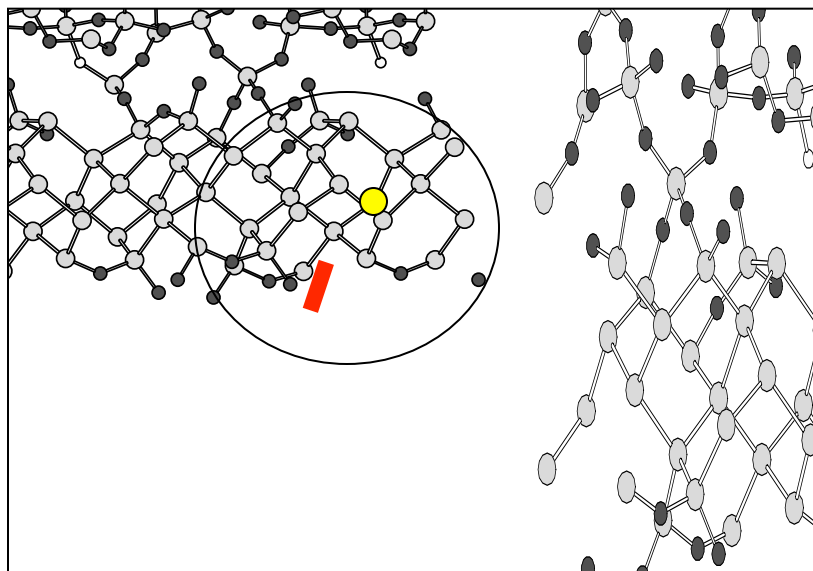
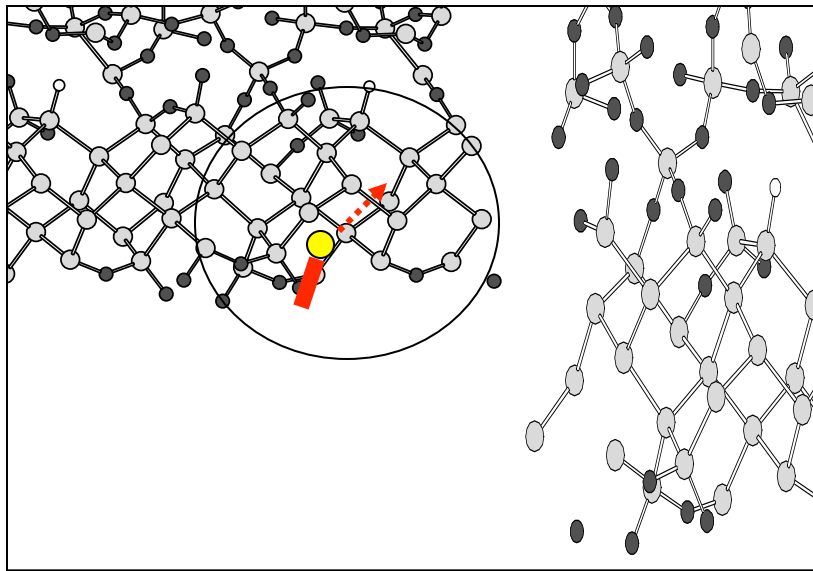
Possible new defect complex

PASSIVATED DB (**Si-H**) ACROSS FROM A SUBOXIDE BOND (Si-Si)

OR Hf-Si BOND

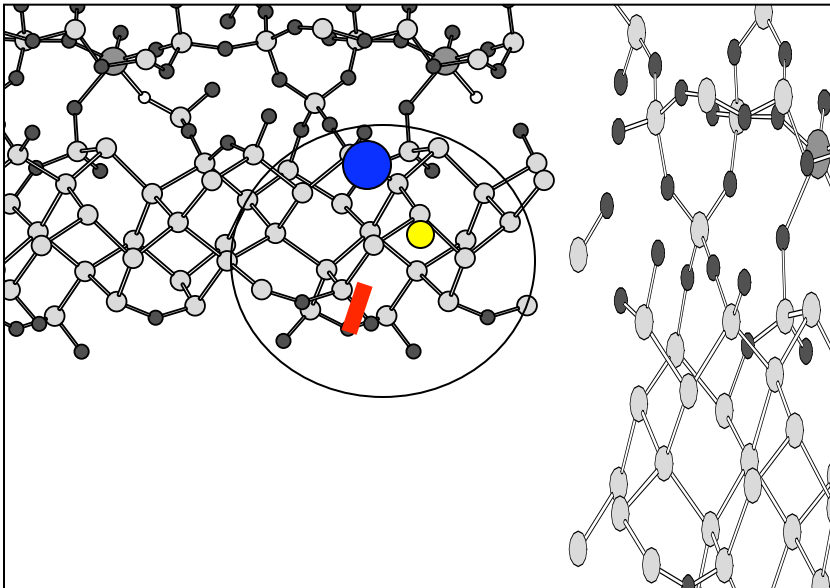
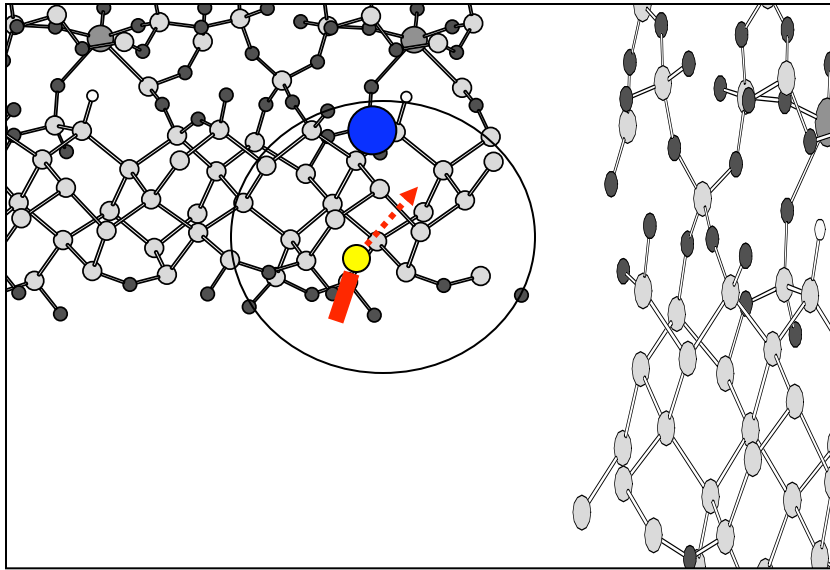


Proton hopping from dangling bond to suboxide bond



BARRIER $>$ 2.2 eV

Alternative: DB ACROSS FROM Hf-Si BOND



BARRIER ~ 1.1 eV

NEW RESULTS

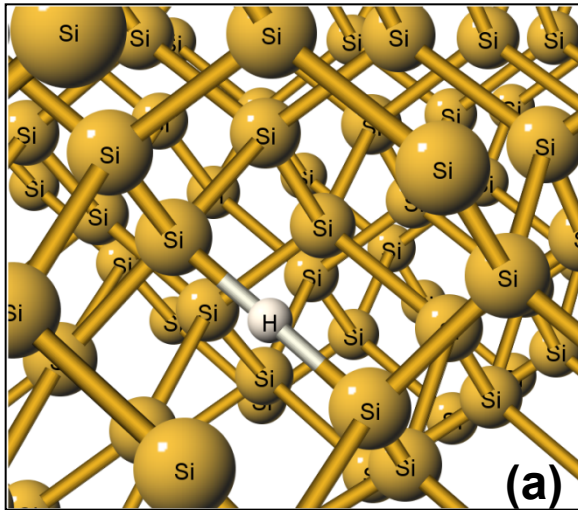
DEFECT DYNAMICS IN SiGe/strained-Si

- FREE HYDROGEN
- PASSIVATION OF DOPANTS BY HYDROGEN
- VACANCIES AND INTERSTITIALS

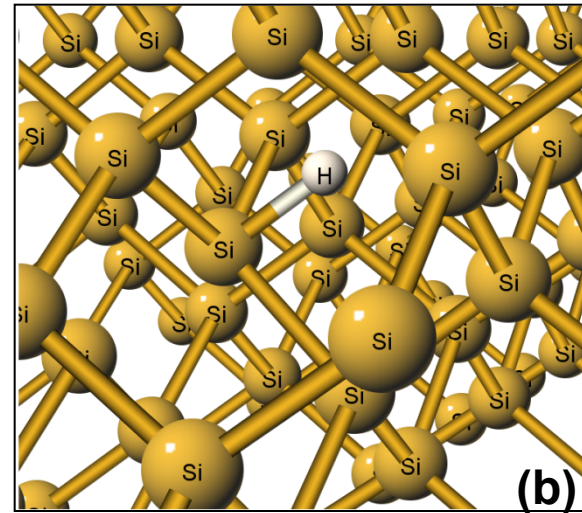
L. Tsetseris, D. M. Fleetwood, R. D. Schrimpf, and S. T. Pantelides,
submitted to Appl. Phys. Lett.

Hydrogen in SiGe and strained Si

H^0, H^+



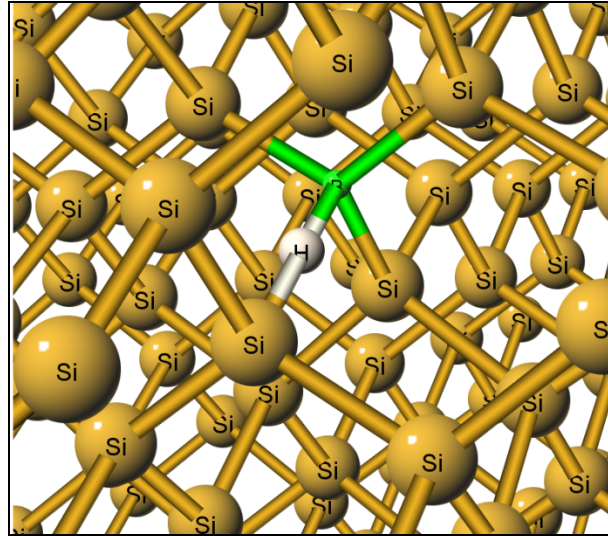
H^-



x_{Ge}	Charge	ΔE (eV)
11.1%	0	0.01
22.2%	0	0.07
33.3%	0	0.05

H^0 prefers SiGe

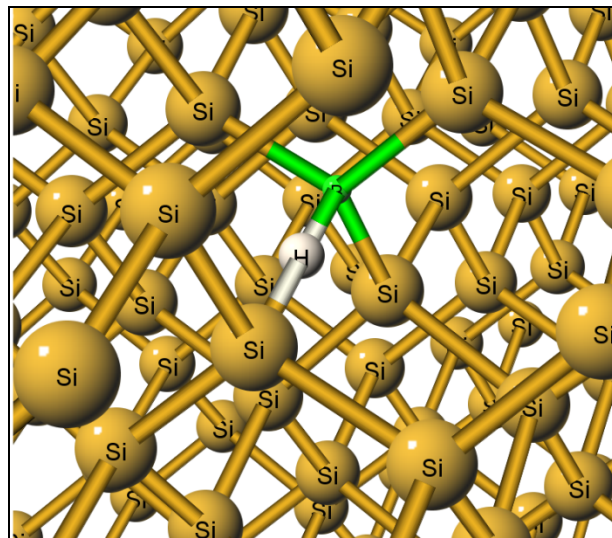
Hydrogen-boron complexes in SiGe and s-Si



x_{Ge}	Charge	E_b (eV)	ΔE (eV)
11.1%	0	0.61	0.00
22.2%	0	0.63	-0.22
33.3%	0	0.57	-0.54

H prefers to stick to B in s-Si.

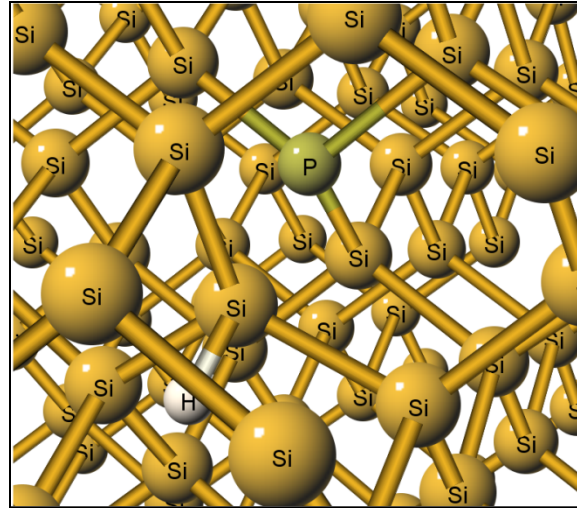
Hydrogen-boron complexes in SiGe and s-Si



x_{Ge}	Charge	E_b (eV)	ΔE (eV)
11.1%	0	0.61	0.00
22.2%	0	0.63	-0.22
33.3%	0	0.57	-0.54
11.1%	-	0.48	0.00
22.2%	-	0.54	-0.08
33.3%	-	0.44	-0.13

H prefers to stick to B in s-Si.

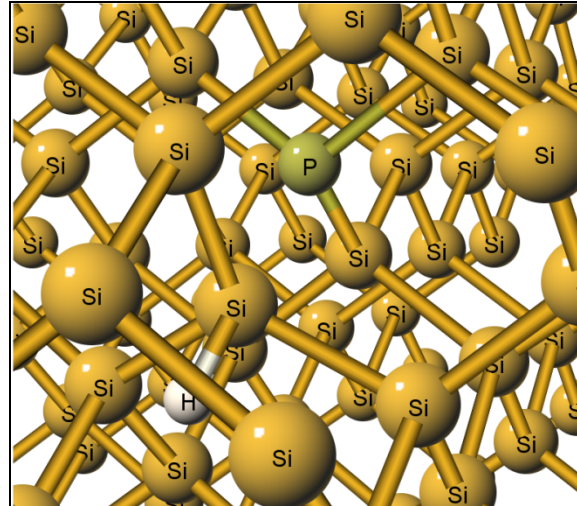
P-H complexes in SiGe and s-Si



x_{Ge}	Charge	E_b (eV)	ΔE (eV)
11.1%	0	0.70	0.00
22.2%	0	0.45	0.18
33.3%	0	0.52	0.42

H prefers to stick to P sites in SiGe.

P-H complexes in SiGe and s-Si



x_{Ge}	Charge	E_b (eV)	ΔE (eV)
11.1%	0	0.70	0.00
22.2%	0	0.45	0.18
33.3%	0	0.52	0.42
11.1%	+	0.31	0.00
22.2%	+	0.27	0.01
33.3%	+	0.30	0.01

H prefers to stick to P sites in SiGe.

IMPACT ON DEVICE OPERATION

NBTI, PBTI

INCREASE IN INTERFACE TRAP DENSITY
BIAS, MODERATE TEMPERATURES ($\sim 150^\circ \text{C}$)

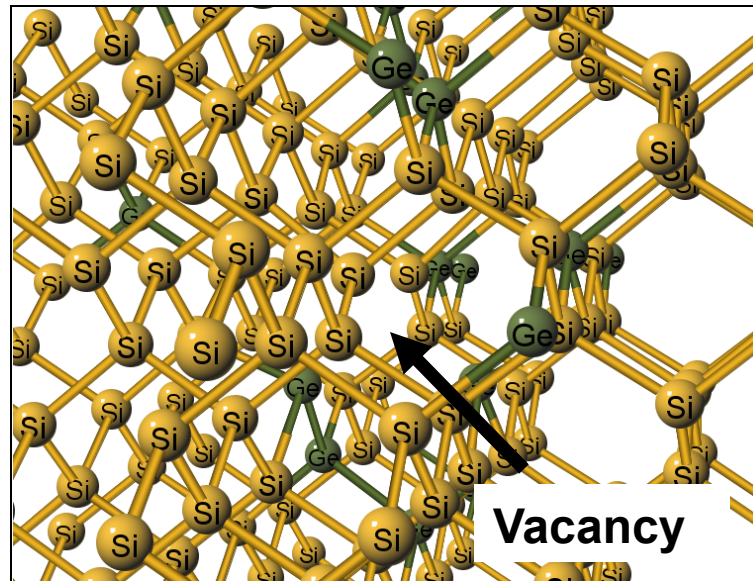
H prefers to stick to B in s-Si

BTI GETS WORSE

H prefers to stick to P sites in SiGe

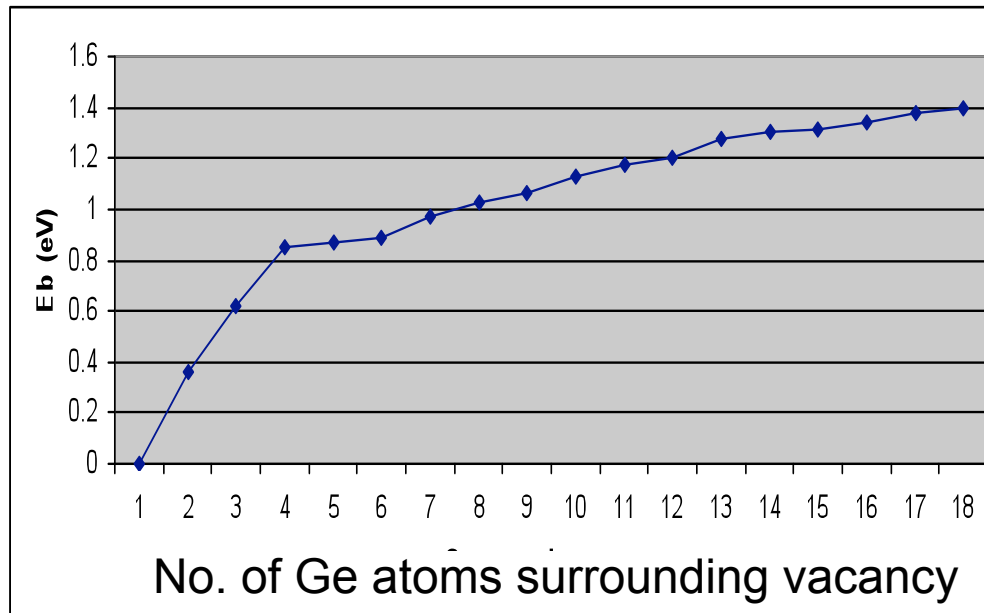
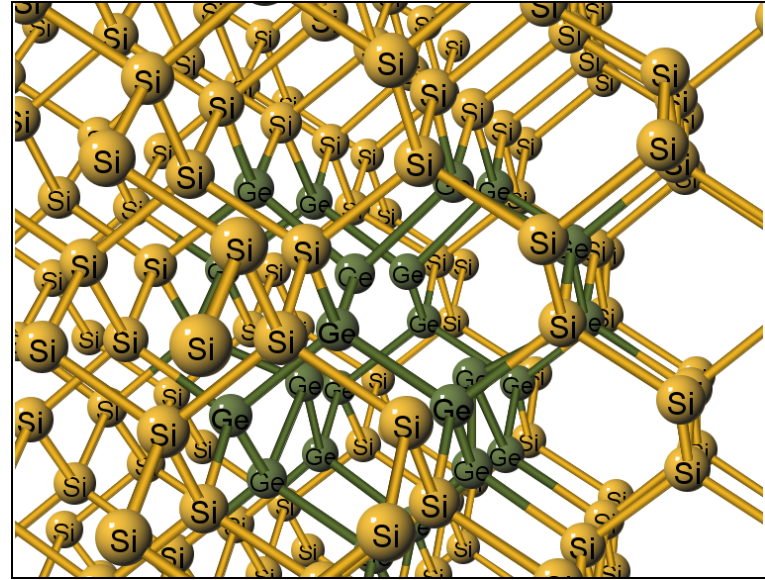
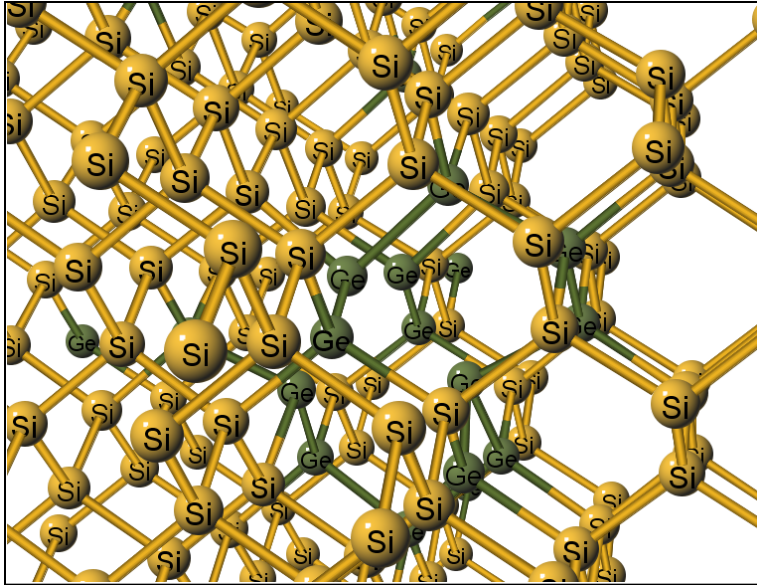
BTI IMPROVEMENT

Vacancies in SiGe/strained Si



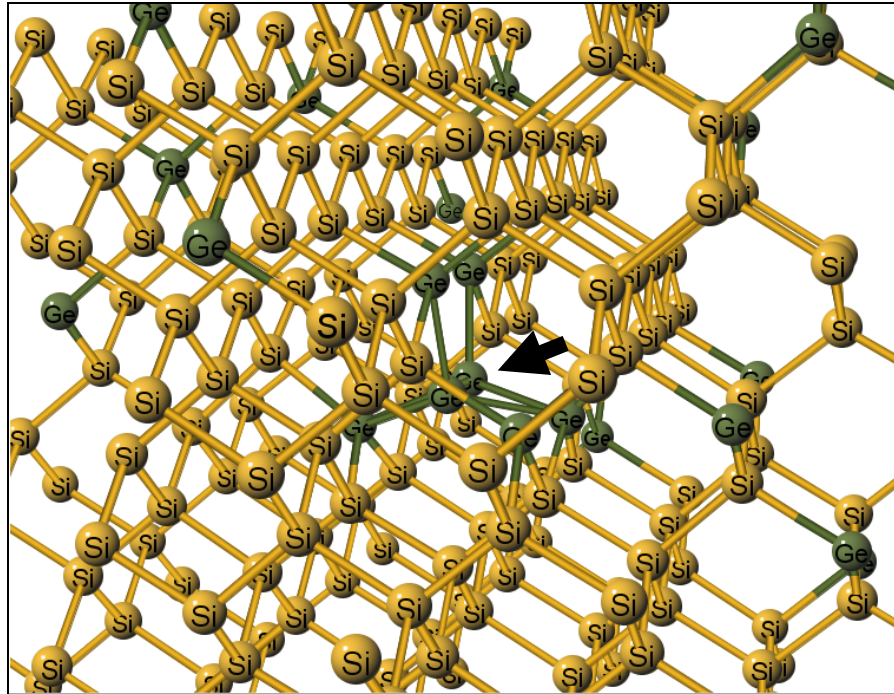
Vacancies have lower energy in SiGe

Agglomeration of Ge atoms around vacancies in SiGe



1.4 eV

Self-interstitials in SiGe/strained Si



- **Si self-interstitial moves from s-Si to SiGe
energy gain 0.23 eV for $x_{\text{Ge}} = 11\%$**
- **Si interstitial goes substitutional, creating a Ge interstitial**
- **Ge atoms agglomerate around Ge self-interstitial
(trend weaker than that for vacancies)**

IMPACT ON DEVICE OPERATION

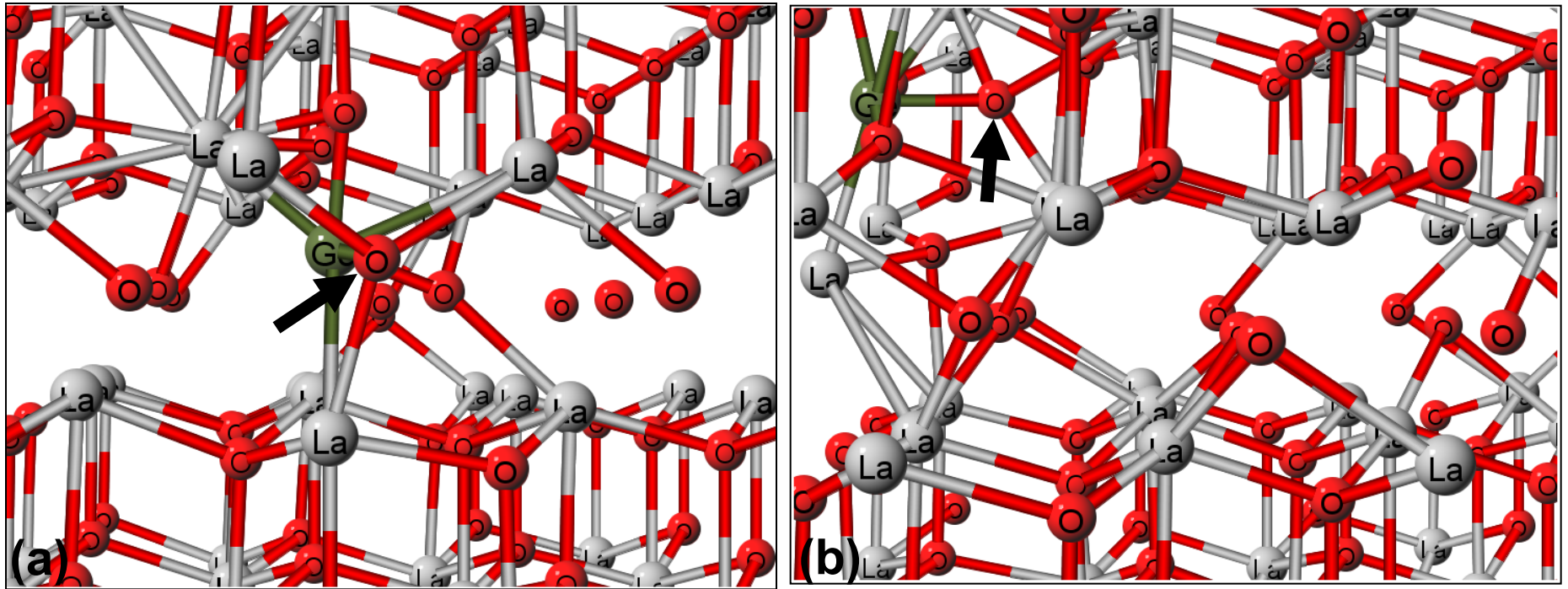
**PRESENCE OF SiGe SUBSTRATE REDUCES
RADIATION-INDUCED DISPLACEMENT DAMAGE**

Ge/high-k systems

Ge volatilization products in high-k dielectrics

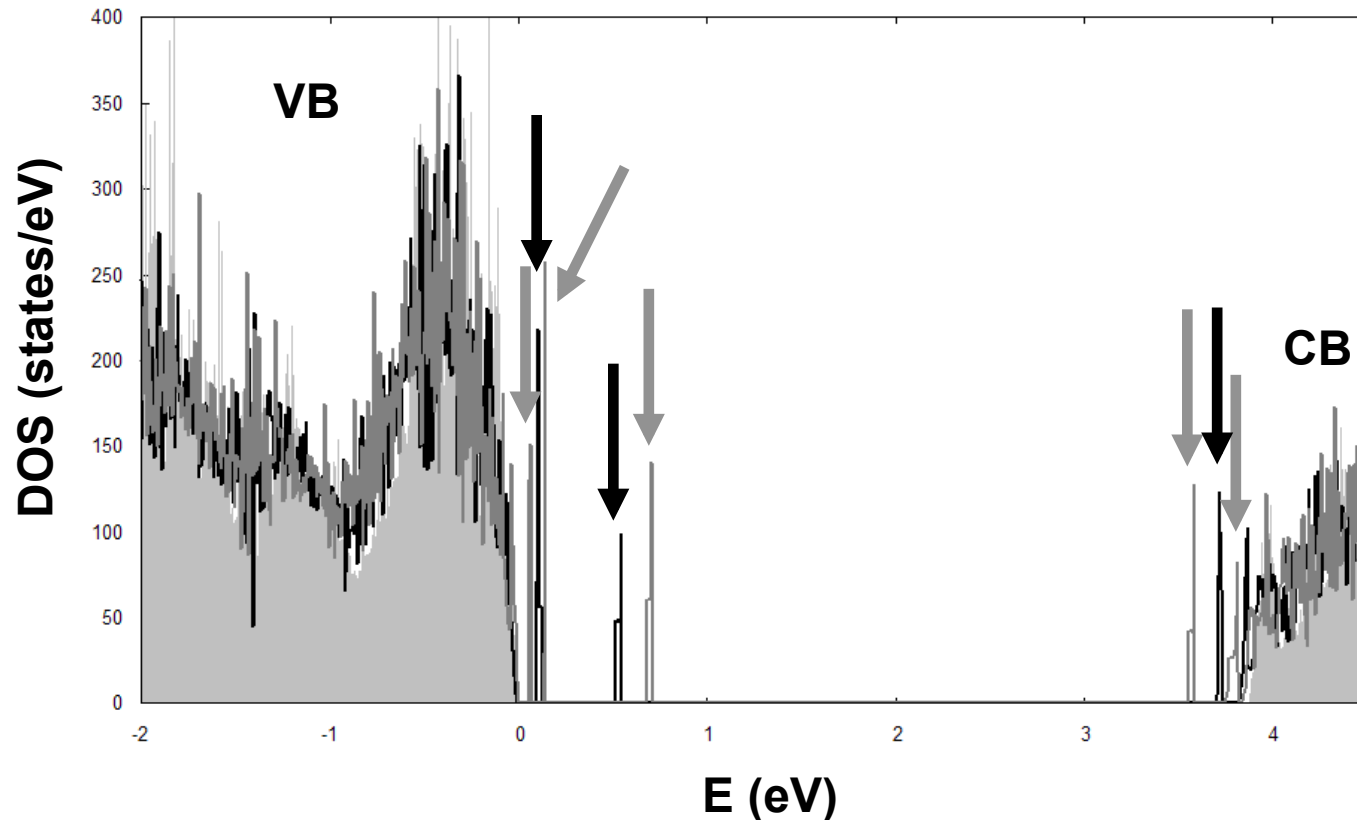
- **Annealing of Ge substrates** initiates volatilization and desorption of GeO molecules
- **In Ge-based gate stacks**, GeO must out-diffuse through the gate dielectric
- **Some of the GeO molecules may be trapped in the gate oxides**
- **Experimental evidence:** out-diffusion of Ge-related species causes degradation of the dielectric (e.g. charge trapping, enhanced leakage current)

GeO impurities in La_2O_3



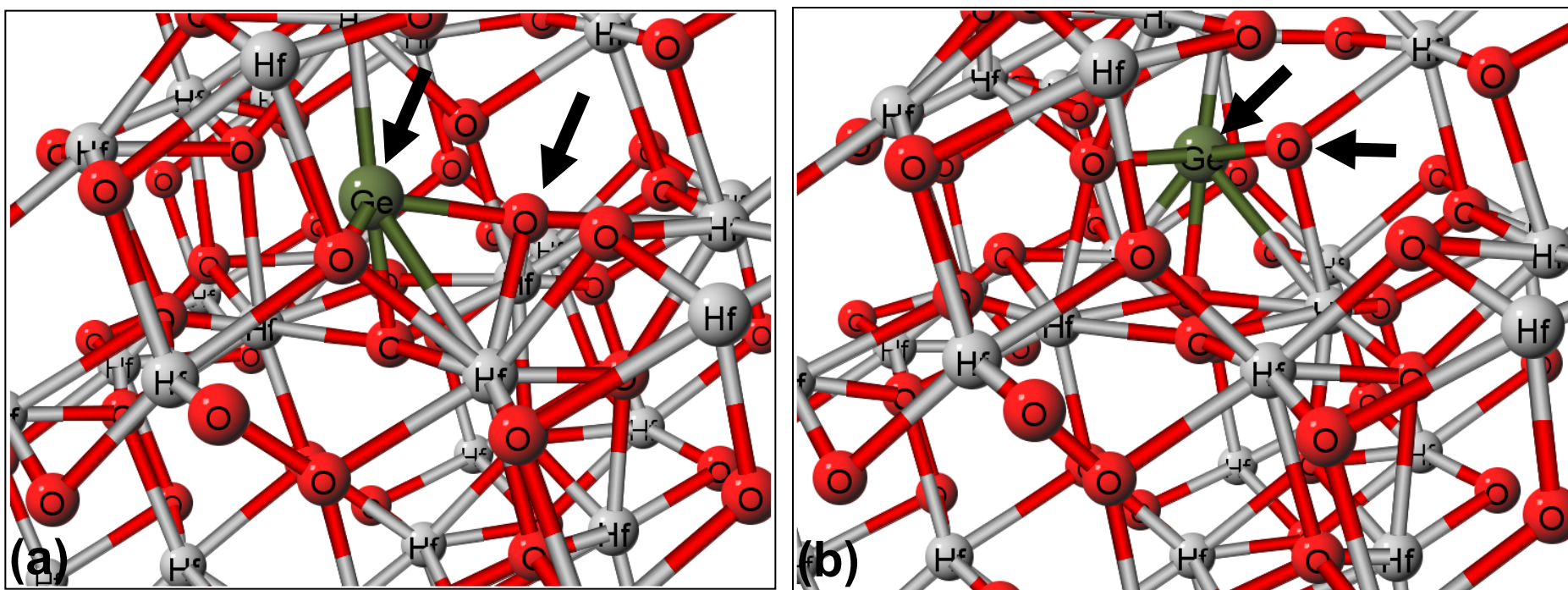
- Hex- La_2O_3 : layered structure with La-rich slices and O atoms in-between.
- Stable configurations of GeO impurities within and inside the layers.
- GeO transfer from La_2O_3 bulk to vacuum: energy gain of 2.7 eV

GeO in La₂O₃: Electronic properties



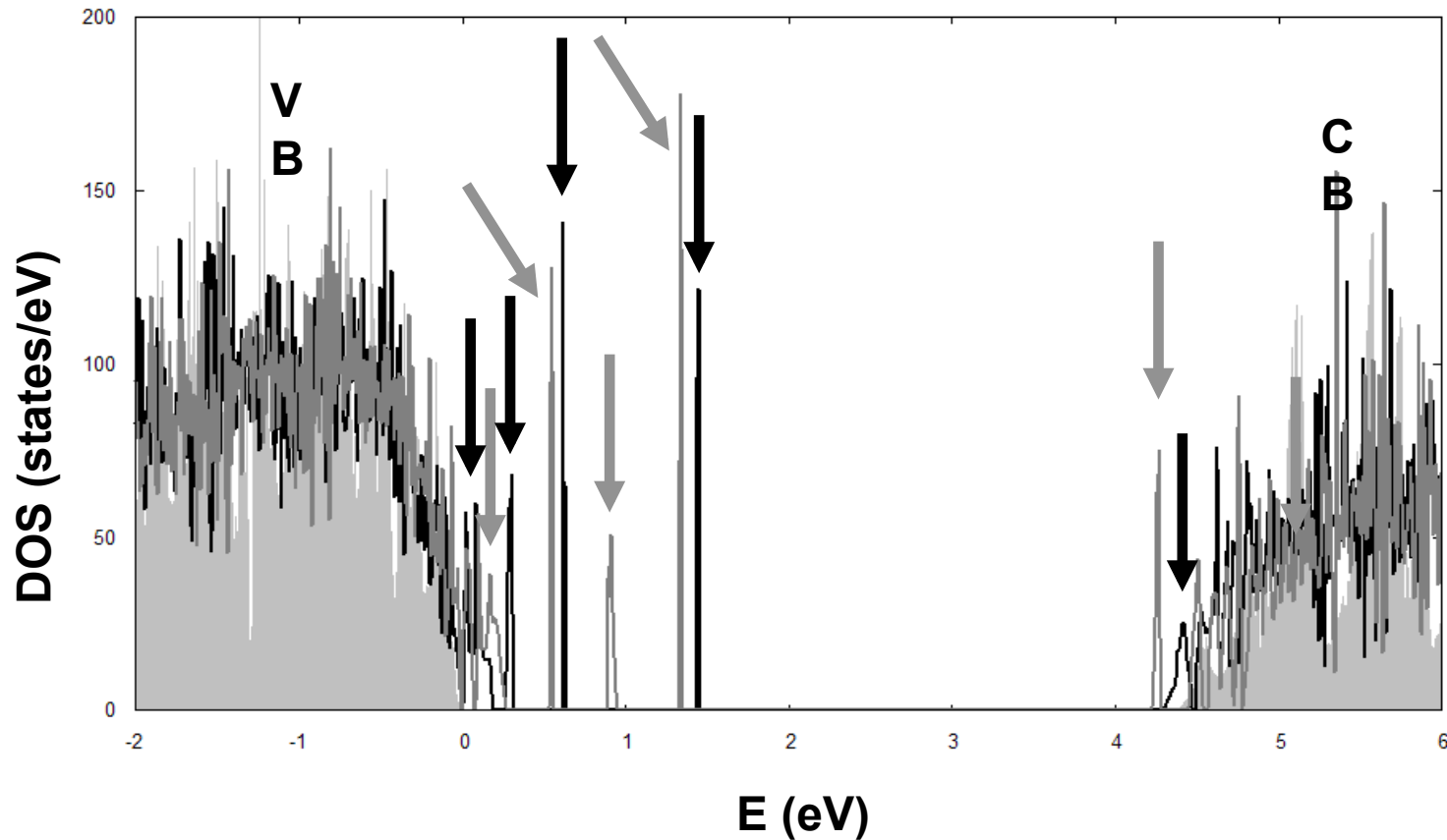
- Energy levels in the gap cause **charge trapping** and enhance **leakage currents**
- Hydrogen binds to GeO impurities, but some levels in the gap remain

GeO impurities in HfO₂



**GeO transfer from HfO₂ bulk to vacuum:
energy gain of 1.5 eV**

GeO in HfO₂: Electronic properties



- Energy levels in the gap cause **charge trapping** and enhance **leakage currents**
- Hydrogen binds to GeO impurities, but some levels in the gap remain

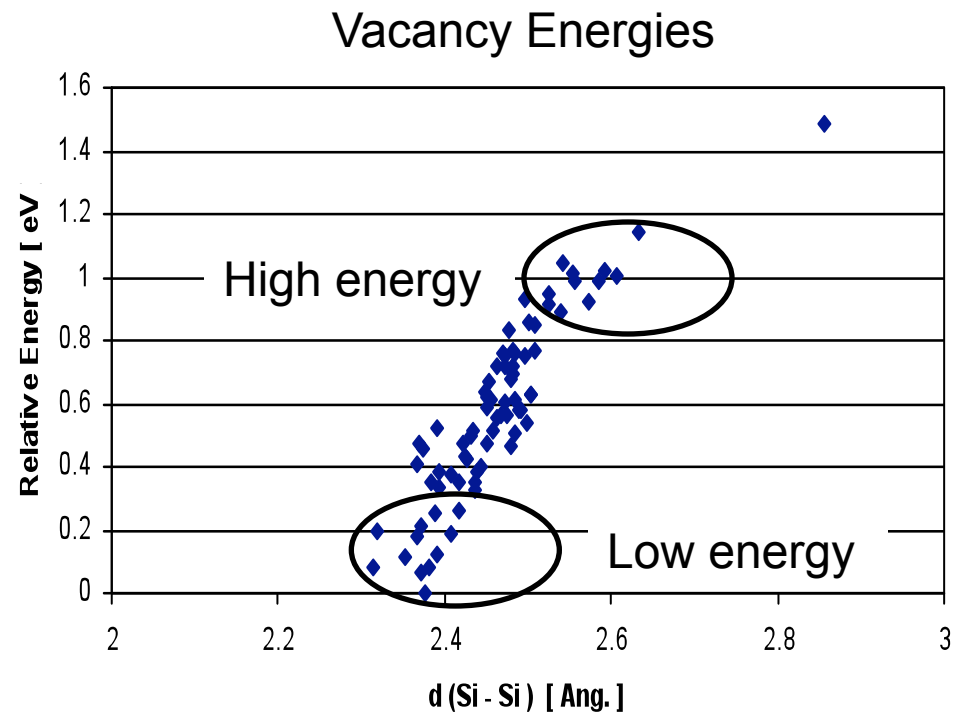
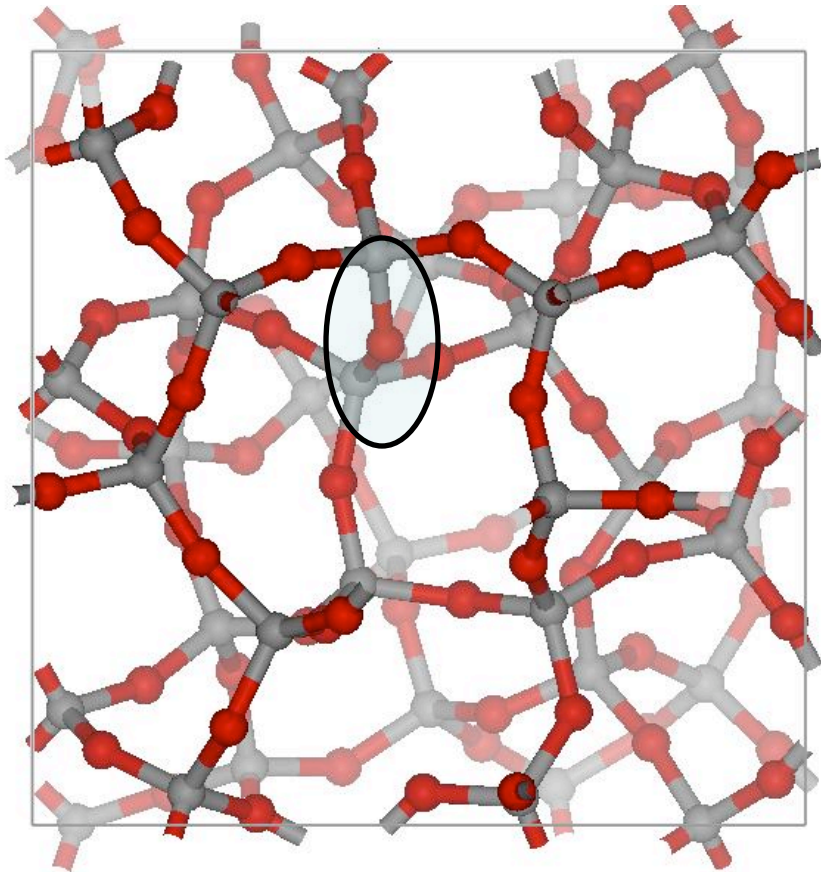
CRACKING OF H₂ IN SiO₂

- Conley and Lenahan (1993):

Experimental evidence in support of H₂ cracking at O vacancies

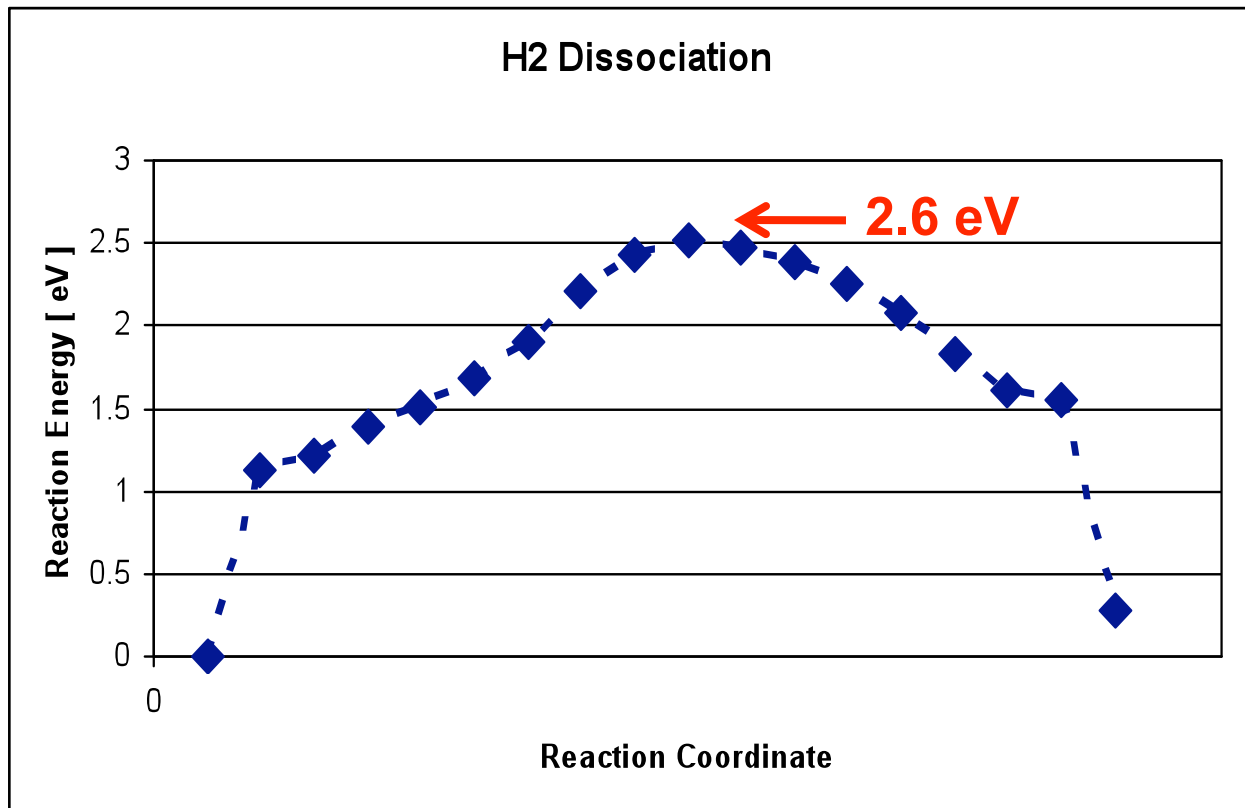
- Contrary to theoretical results
(Edwards and coworkers 1992, 1993)

Oxygen Vacancies

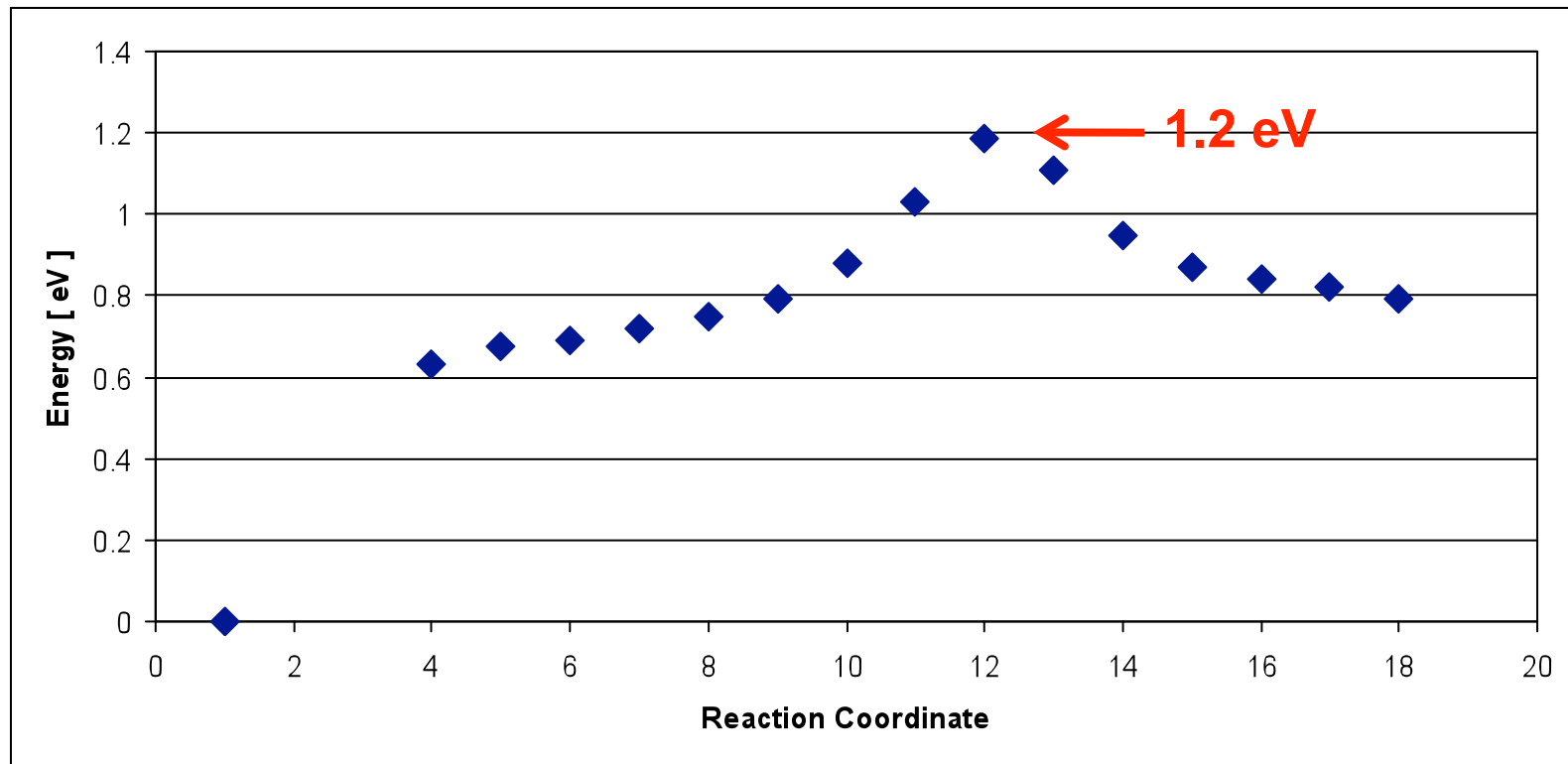


$T \sim 1200 \text{ K}$
 $N_L / N_H \sim e^{1\text{eV}/kT} \sim 10^4$

H₂ reacting at V

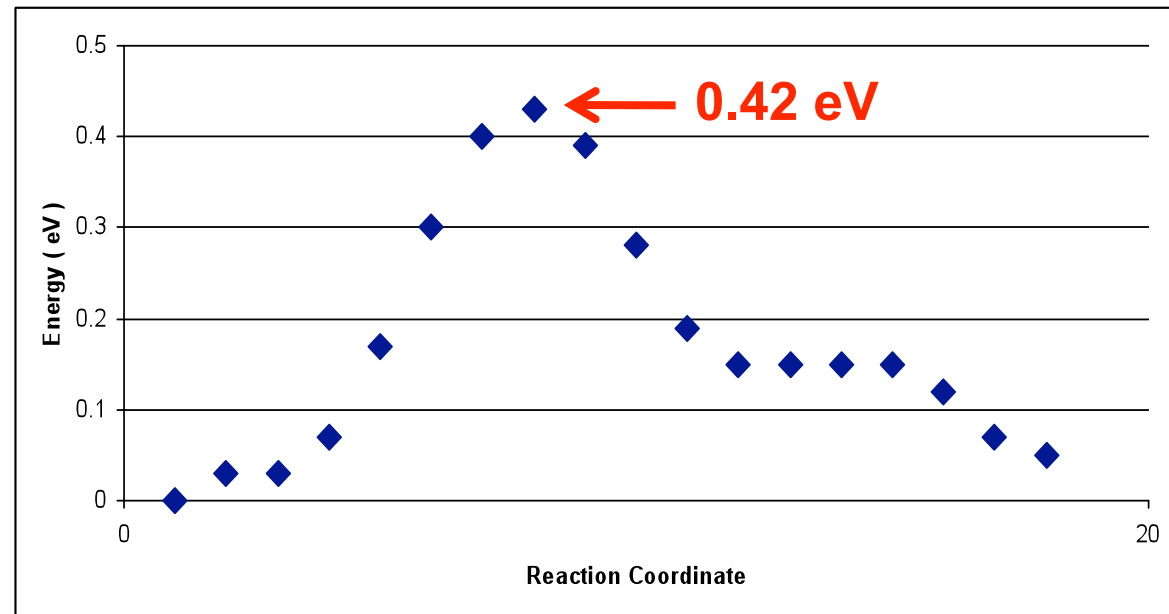
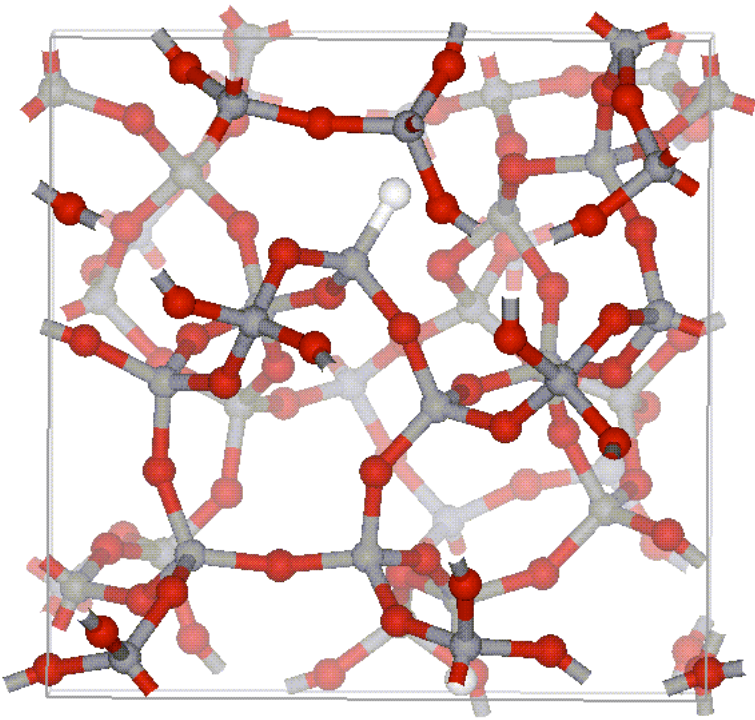
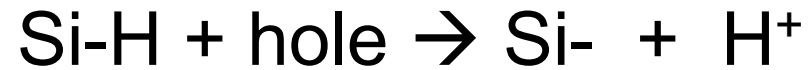


H₂ reacting at V⁺



H₂ does not crack at room temperature at low-energy vacancies!

ALTERNATIVE MECHANISM: Si DANGLING BONDS



**H₂ cracks at a bare dangling bond
without an energy barrier**