

June 10 - Annual Muri Review

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Additional Support: NSF, SRC and DTRA

outline

- i) mono- and di-vacancy defects in HfO₂**
- ii) monoclinic HfO₂ and HfSiON on Ge**

**objective - provide a science base for optimization
of HfO₂ and Hi Si oxynitride dielectrics
for survivable devices, rad hard on Si and Ge substrates**

**X-ray absorption and photoemission spectroscopies, XAS
and SXPS,**

conduction and valence band electronic states

band edge, intrinsic and chemical defects

**defects, and dependence on processing are then compared
with defects, traps and fixed charge, detected electrically**

relationship to

CMOS

(i) define and narrow the field of high-k dielectrics that meet the aggressive scaling required deep

sub-micron- and nano-CMOS, and

(ii) identify processing constraints that apply to

Ge and Si substrates

rad testing

(i) differences in rad hardness/survivability of generically similar dielectrics

HfO₂ and Hf Si oxynitride on Si and Ge substrates

and correlations with

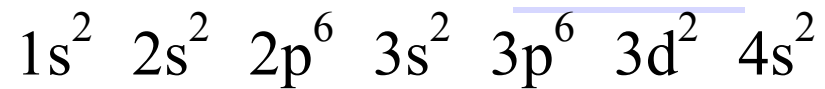
DEFECTS IN DIELECTRICS

- 1. Why do we care? Leakage paths, breakdown points, trapping centers in radiation damage. Mostly oxygen vacancies**
- 2. Nature of defects----intrinsic or processing dependent ?**
- 3. Vacancies are hard to “see”---positrons, energetic electron displacement, material contraction. This work-x-ray, electron spectroscopy!**
- 4. Oxygen vacancies plague all of the current oxide work!**

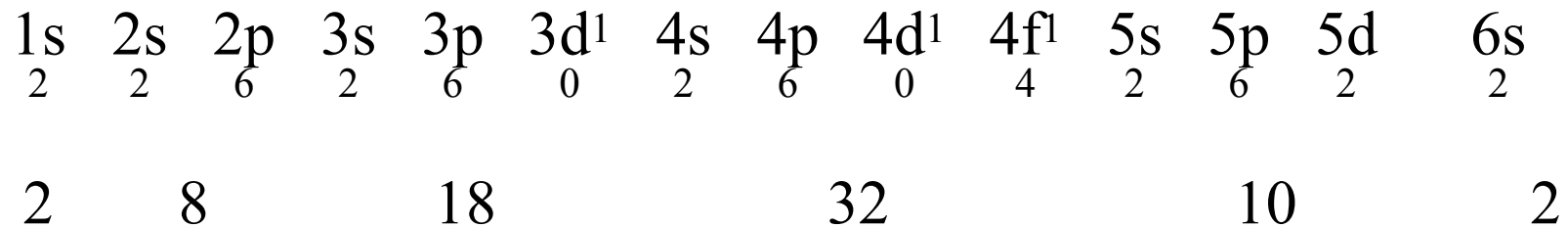
H 1																	He 2
Li 3	Be 4											B 5	C 6	N 7	O 8	F 9	Ne 10
Na 11	Mg 12											Al 13	Si 14	P 15	S 16	Cl 17	Ar 18
K 19	Ca 20	Sc 21	Ti 22	V 23	Cr 24	Mn 25	Fe 26	Co 27	Ni 28	Cu 29	Zn 30	Ga 31	Ge 32	As 33	Se 34	Br 35	Kr 36
Rb 37	Sr 38	Y 39	Zr 40	Nb 41	Mo 42	Tc 43	Ru 44	Rh 45	Pd 46	Ag 47	Cd 48	In 49	Sn 50	Sb 51	Te 52	I 53	Xe 54
Cs 55	Ba 56	Lu 71	Hf 72	Ta 73	W 74	Re 75	Os 76	Ir 77	Pt 78	Au 79	Hg 80	Tl 81	Pb 82	Bi 83	Po 84	At 85	Rn 86
Fr 87	Ra 88	Lr 103	Db 104	Jl 105	Rf 106	Bh 107	Hn 108	Mt 109									

La 57	Ce 58	Pr 59	Nd 60	Pm 61	Sm 62	Eu 63	Gd 64	Tb 65	Dy 66	Ho 67	Er 68	Tm 69	Yb 70
Ac 89	Th 90	Pa 91	U 92	Np 93	Pu 94	Am 95	Cm 96	Bk 97	Cf 98	Es 99	Fm 100	Md 101	No 102

22 Ti titanium : [Ar] 3d² 4s²



72 Hf hafnium : [Xe] 4f¹⁴ 5d² 6s²



generate, E_g , groups as determined by the coordination and local symmetry of the atoms with respect to their O-atom neighbors, and then (2) the removal of these twofold degeneracies by Jahn-Teller (J-T) bonding distortions that are accompanied by symmetry reductions at the TM atom bonding sites [12,20]. See Figure 10.1 for a schematic

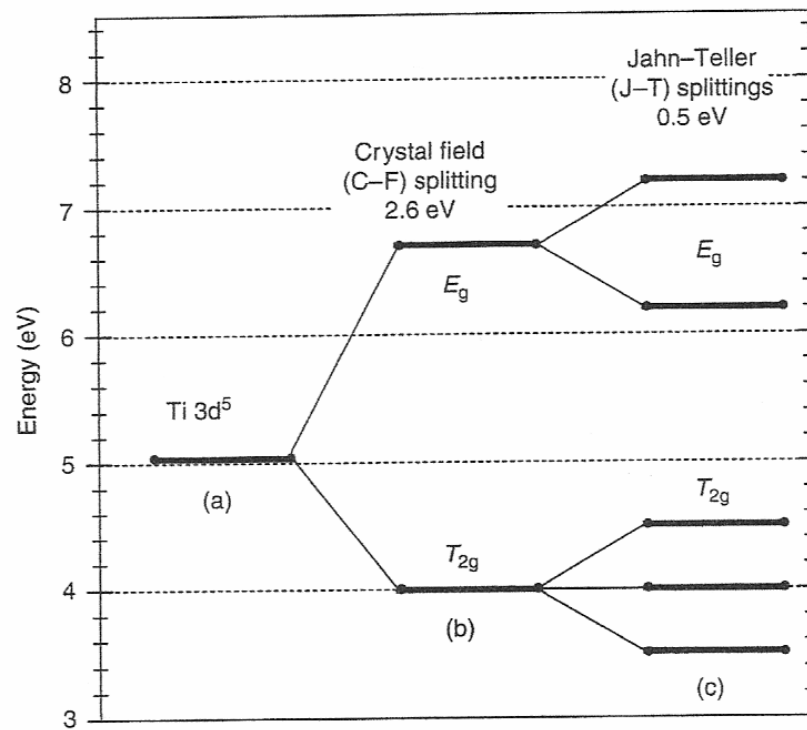


FIGURE 10.1

Schematic representation of C-F and Jahn-Teller splittings for Ti relative to the fivefold degenerate atomic state: (a) the atomic Ti 3d state; (b) in an ideal octahedral field; and (c) in a distorted octahedral field. The energy scale is arbitrary, i.e., the energy of the triply degenerate T_{2g} state has been set to 4 eV.

electronic structure of transition metal oxides - FA Cotton

bonding and anti-bonding states - symmetry adopted linear combinations of atomic states/symmetries of constituent atoms

ionic model

valence band occupied O 2p σ and π conduction band empty states Ti 3d 4s, 4p for 6-fold coordination TiO₂
empty states Hf 5d 6s, 6p, 5f for 7/8-fold coordination HfO₂

conduction band empty states not restricted to atomic valence state

valence states - O²⁻ O 2s², 2p²

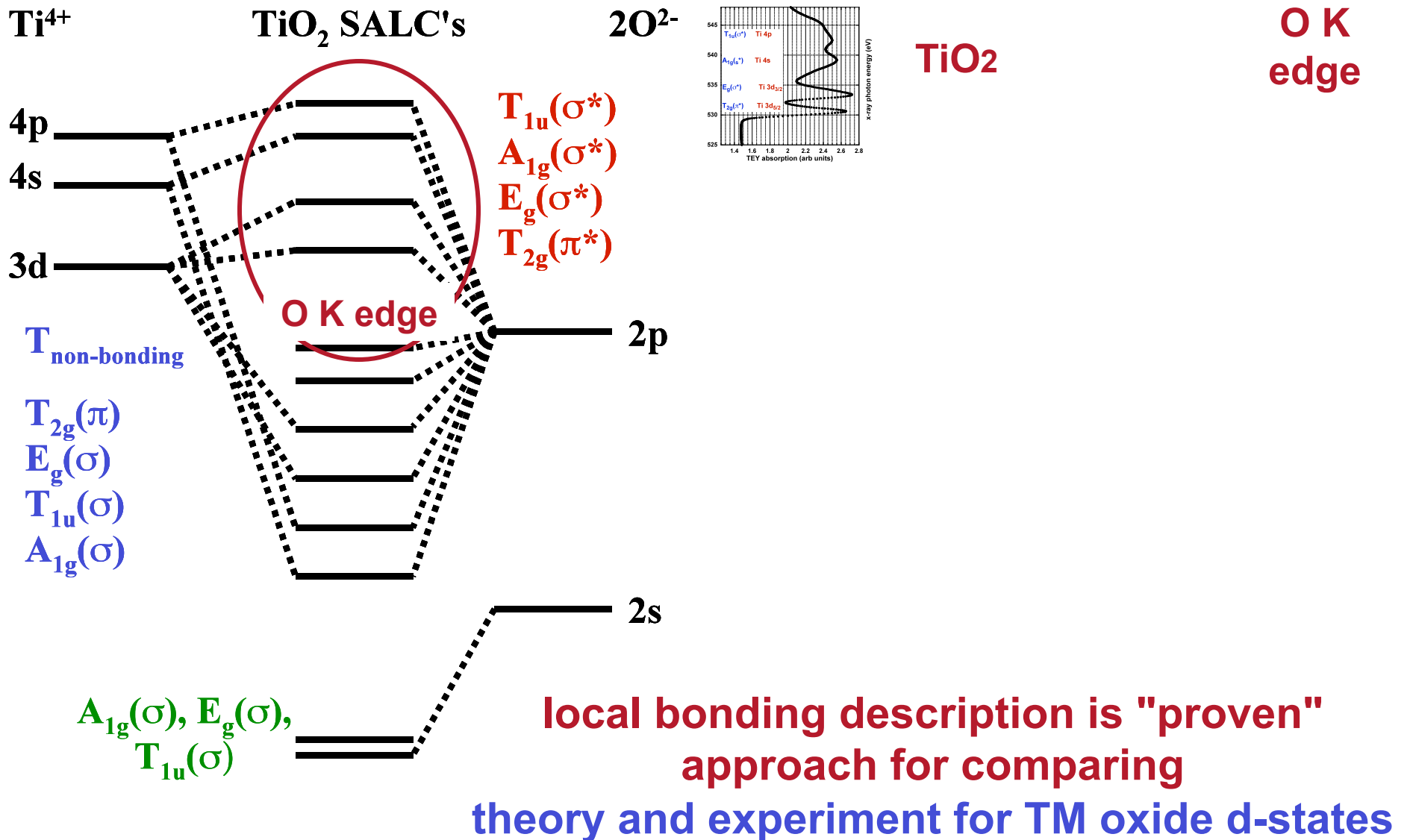
Ti⁴⁺ Ti 3s², 3p⁶ d⁰ configuration, no occupied d-states

Hf⁴⁺ Hf 5s², 5p⁶ d⁰ configuration, no occupied d-states

partially ionic model
bonding and anti-bonding states - symmetry adopted linear combinations of atomic states/symmetries of constituent atoms
valence-conduction band transitions - O 2p - O 2p*, each mixed with either Ti or Hf states

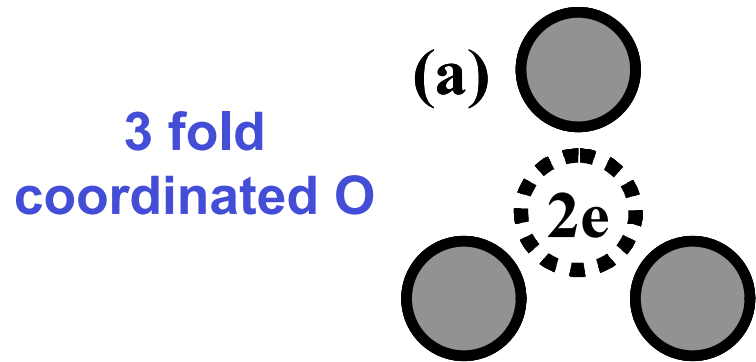
schematic molecular orbital diagram for TiO₂

6 fold-coordinated Ti⁴⁺ ions, and 3 fold-coordinated O²⁻ ions
 covalent mixing of Ti and O atomic states in symmetry adapted
 molecular orbital states, SALCs - FA Cotton, 1962

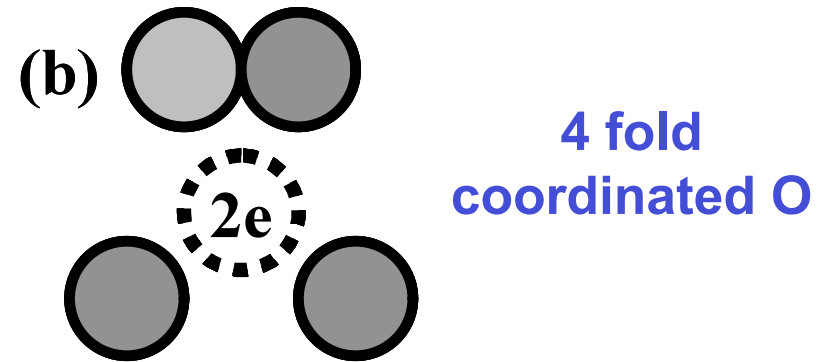


schematic representation of local bonding in monovacancies and divacancies

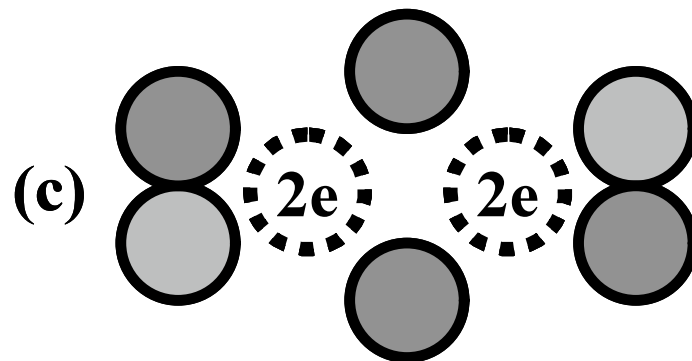
(a) mono-vacancy defect in TiO_2



(b) monovacancy defect in HfO_2



equivalent to two edge connected tetrahedra



distorted octahedron with 4 electrons

(c) divacancy defect in HfO_2

theory-supported model for vacancy defects

charge neutrality each O-atom - replaced by 2 electrons

**Robertson/Shluger distribute these electrons over
d-states of all nearest-neighbor Hf or Ti atoms:**

theory - d-states, not contributing to bonding, are occupied digitally

**d^n representation - n is the number of d-electrons
not contributing to bonding**

determined by formal valence of TM in ionic model

HfO₂ Hf atom valence electrons - $6s^25d^2$

TiO₂ Ti atom valence electrons - $4s^23d^2$

Hf⁴⁺, Ti⁴⁺ are d^0 after 4 electrons are removed

no occupied states above VB edge

defect states - equivalent to pairs of T³⁺ - $4s^23d^1$

with one occupied d-state - $2d^1$ or $1d^2$

(mono)-vacancy defects for TiO₂ are d^2

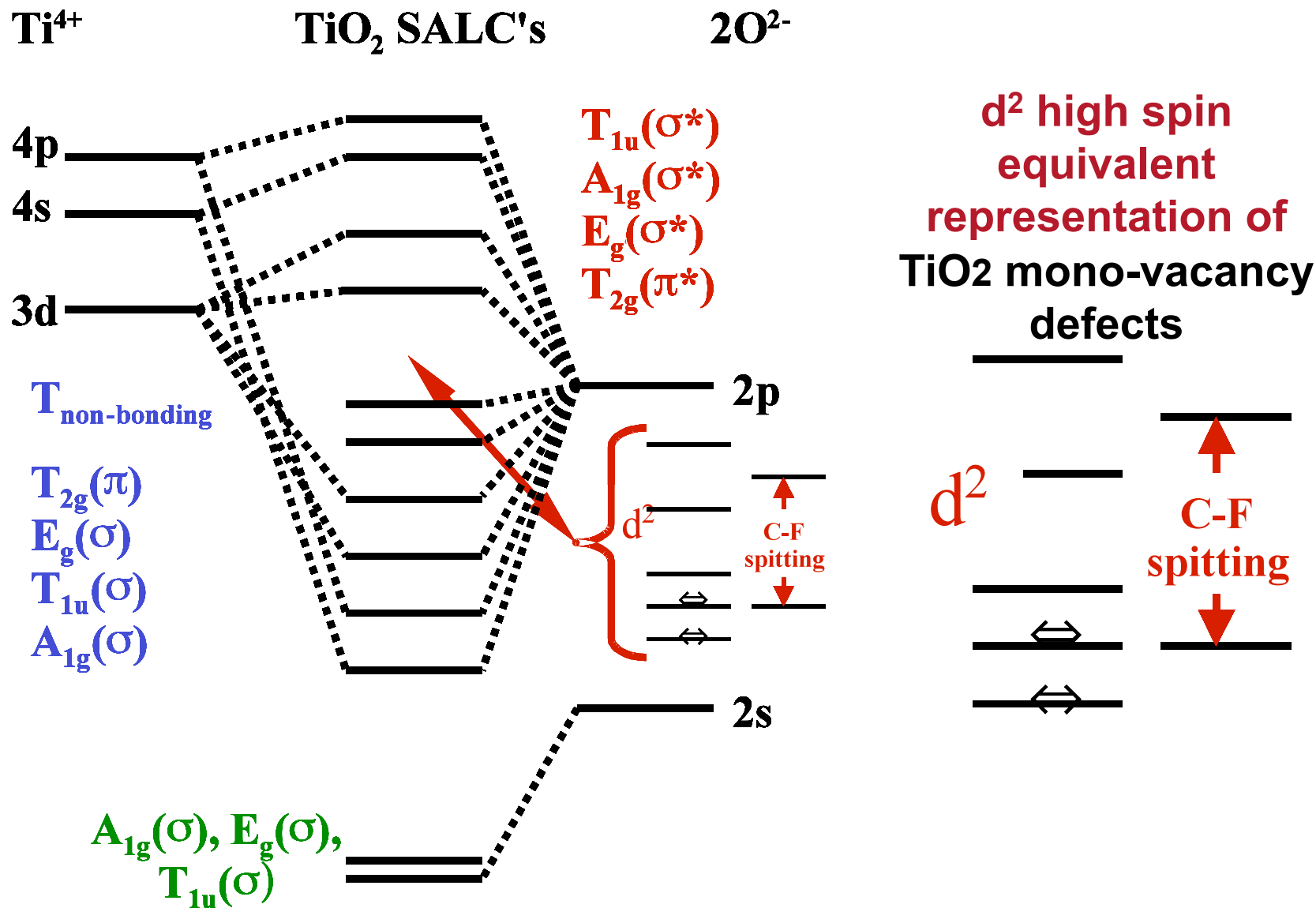
2 occupied states at valence band edge

divacancy defects for HfO₂ are d^4

defect states for d^2 "fit" into band gap: TiO_2

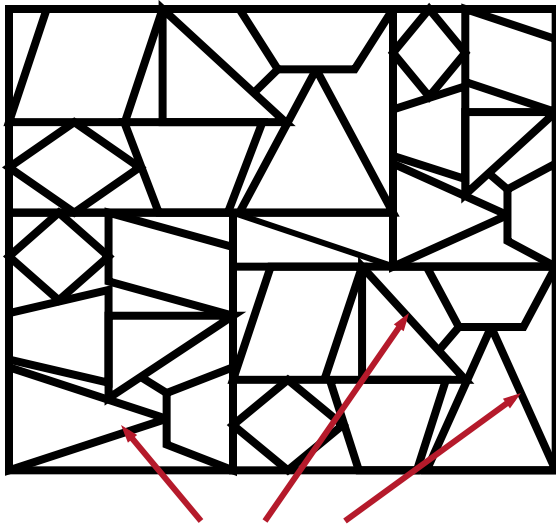
$E_{\text{gap}} = 3.1 \text{ eV} > \Delta(\text{C-F}) = 2.6 \pm 0.1 \text{ eV}$

HfO_2 $E_{\text{gap}} = 5.7 \text{ eV} > \Delta(\text{C-F}) = 3.6 \pm 0.2 \text{ eV}$



Micro-structure of nano-crystalline HfO₂ and non-crystalline Hf silicates and nitrided Hf silicates

nano-crystalline HfO₂



grain boundaries

crystallite size: 2 to 100 nm
grain boundaries - intrinsic bonding defects
local defects at band edges
trap both holes and electrons

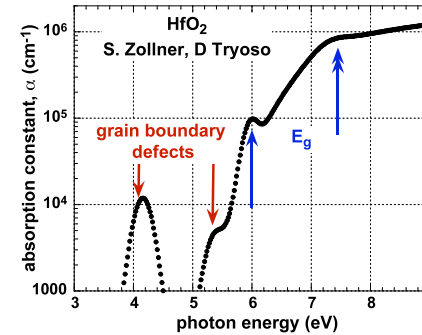
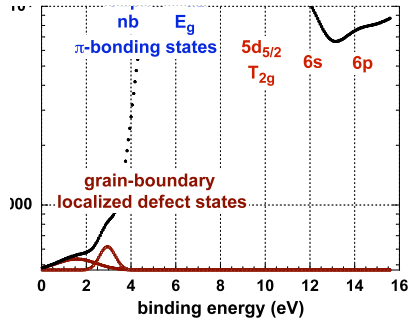
amorphous Hf silicate



isotropic and homogeneous
no discernable granularity
band tail states due to random local atomic structure
localized defect states
generally deep in gap
specific bonding defects
broken bonds - chemical impurities

Spectroscopic evidence for band edge defects in HfO₂

Soft x-ray photoelectron spectroscopy (SXPS)



partially-filled defect states at
valence band edge

hole or electron traps

empty defect states at
conduction band edge

electron traps

**intrinsic defects in elemental and complex TM oxides
have occupied valence band edge states**



rules out Robertson and Shluger mono-vacancies models

K. Xiong, J. Robertson, S.J. Clark, APL 87 (2005)

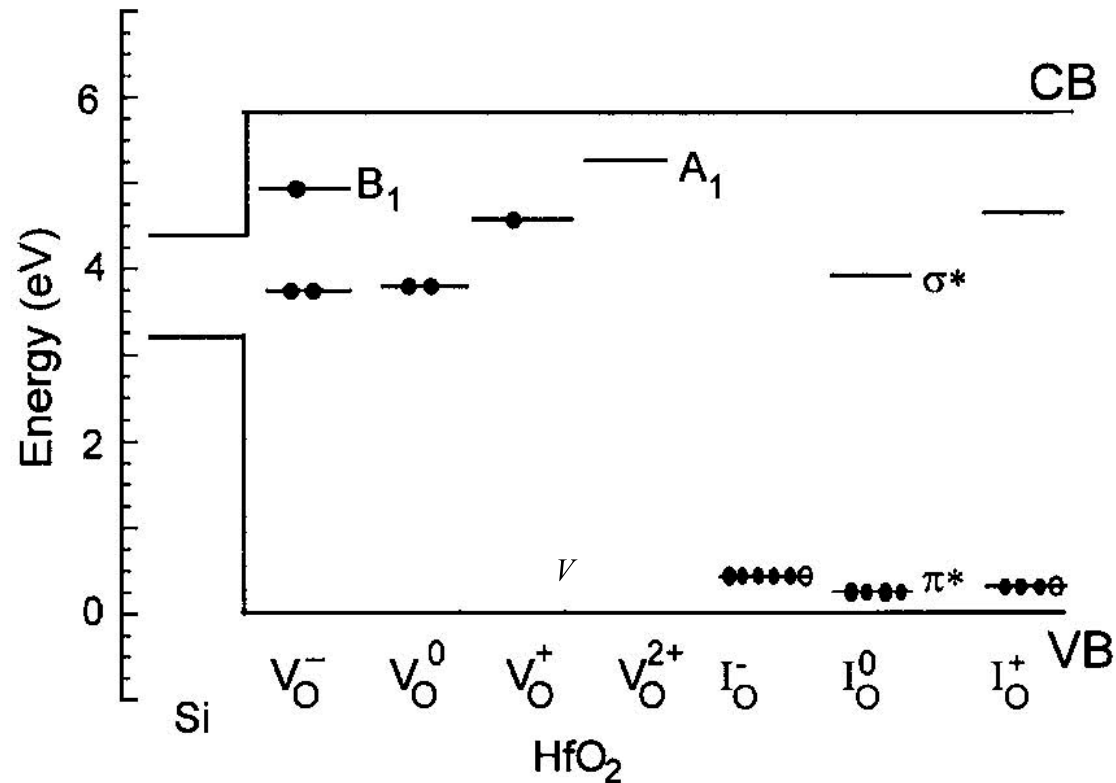


FIG. 2. Summary of calculated energy levels of the relaxed O vacancy and interstitial, in their various charge states.

[V_O]⁻ and [V_O]⁰ are the intrinsic bonding defects
Robertson suggests are identified spectroscopically
and in electrical measurements, but does not explain valence
band edge defects

intrinsic defects in elemental and complex TM oxides are O-atom vacancies

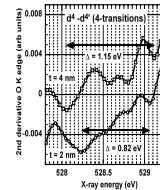
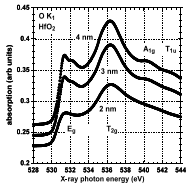
**two issues -- (i) single atom vacancies - mono-vacancies, or
double atom vacancies di-vacancies as in c-Si
(ii) mobile, or fixed or immobile
both resolved spectroscopically**



**occupied defects at valence band edge
rules out Robertson and Shluger mono-vacancies models**

defect density is a function of film thickness - correlates with splitting of E_g d-state feature

larger grains in 4 nm thick film- defect density 10x higher than 2 nm thick film

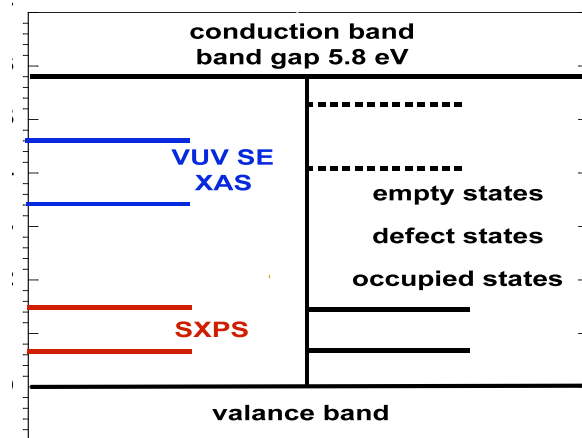


fewer and qualitatively different defects in 2 nm film
mobile defects less effective in holding charge
than immobile defects

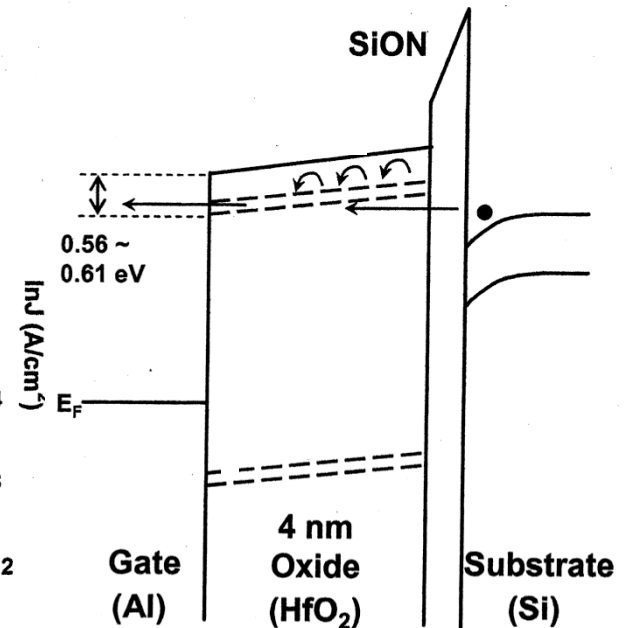
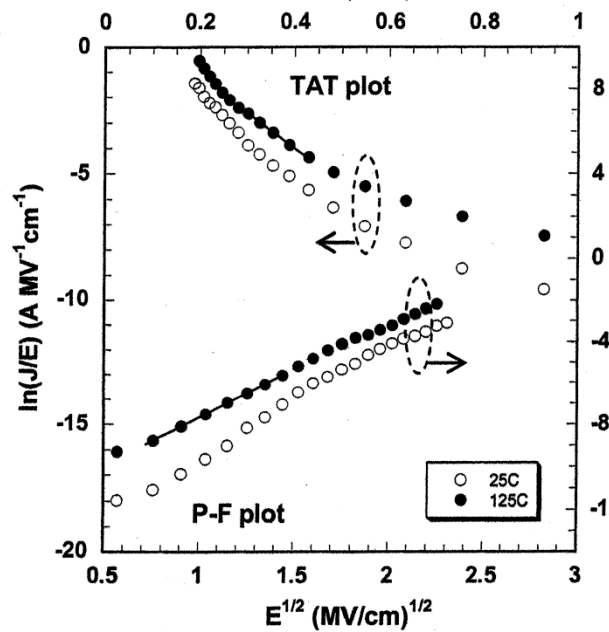
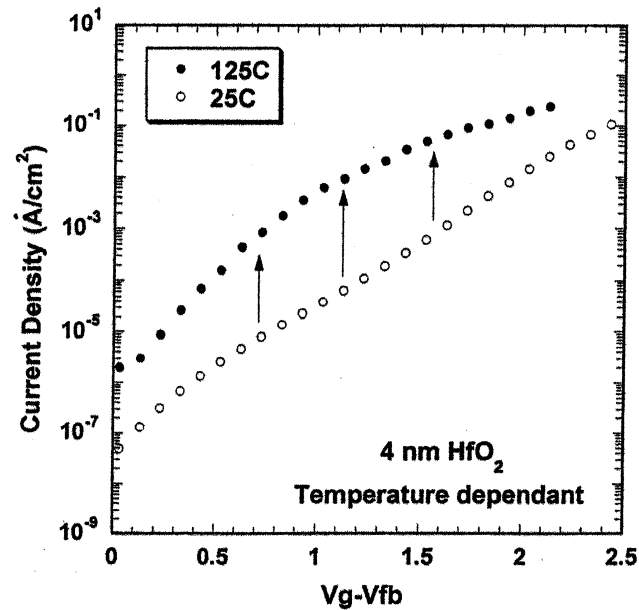
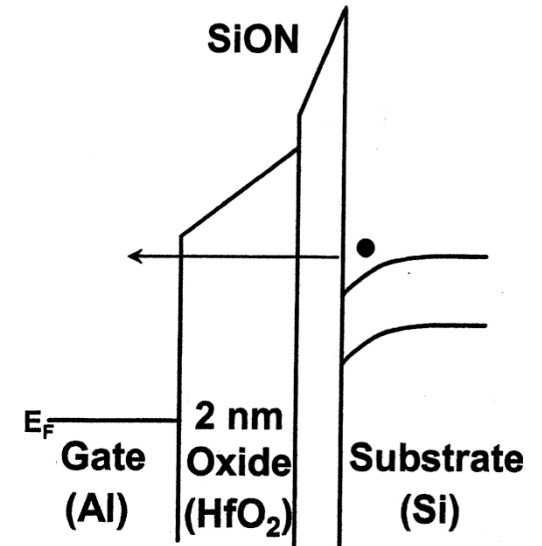
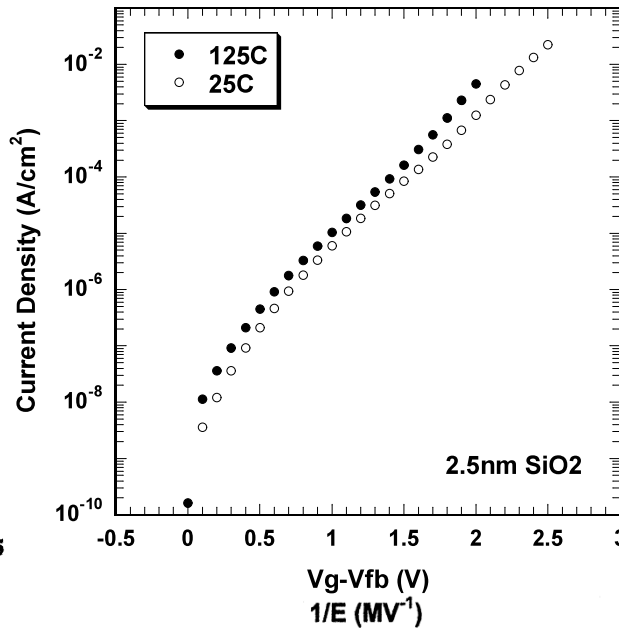
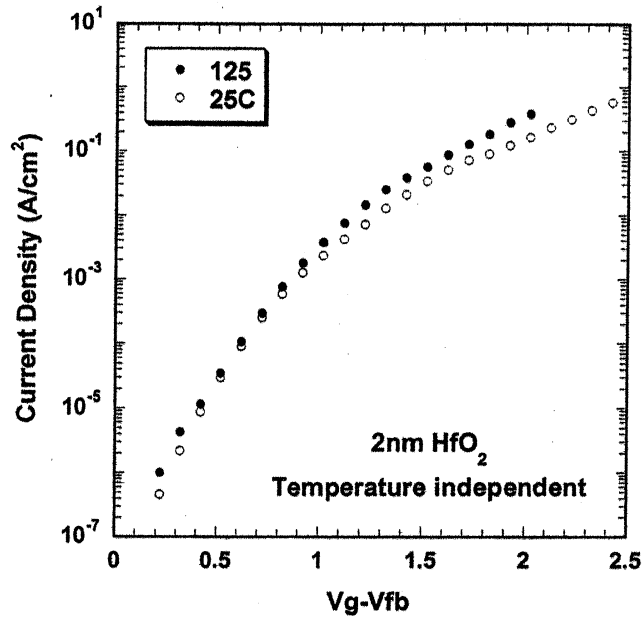
energy level diagram for occupied and empty defects states:

(left) spectroscopic energies referenced to valence band edge, WRONG!

(right) empty state energies referenced to d-d' transitions from occupied states above valence band



differences in J-V correlate with differences in conduction band edge defect states



- **removal of Ge dielectrics between Ge and HfO₂**
- textured epitaxial alignment between HfO₂ and Ge(001)**
- **old clean - acidic -- Hf/H₂O 6 cycles - 'thick' GeO₂**
 - **new clean - neutral/basic - 'thinner' GeO₂**
 - **electrical results**

dielectrics on Ge a “devious” pathway to
eliminate Ge-O and Ge-N bonding

surface nitridation
GeOx - plasma or chemical
Ge (001) substrate

pre-deposition Ge passivation
to prevent active plasma O/Hf/Ti
from reacting with
at Ge surface during deposition

remote plasma HfO ₂ /TiO ₂
surface nitridation
GeOx - plasma or chemical
Ge (001) substrate

remote plasma deposition of
HfO₂/TiO₂ at 300°C
Ge passivation
prevents active plasma O/Hf/Ti
from reacting with
at Ge surface during deposition

chemical = pre-deposition cleans

- old clean - acidic -- Hf/H₂O 6 cycles - ‘thick’ GeO₂
- new clean - neutral/basic - ‘thinner’ GeO₂

post deposition annealing in Ar ambient

(i) two step 550°C / 800°C or (ii) single step 800°C

remote plasma HfO ₂ /TiO ₂
nano-grain growth - textured film
epi-regrowth Ge
Ge (001) substrate

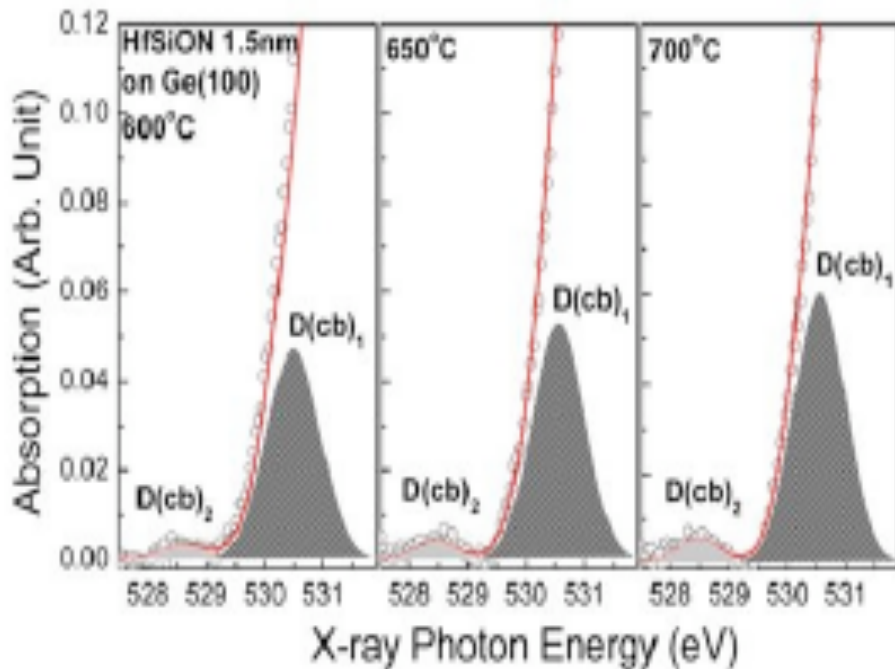
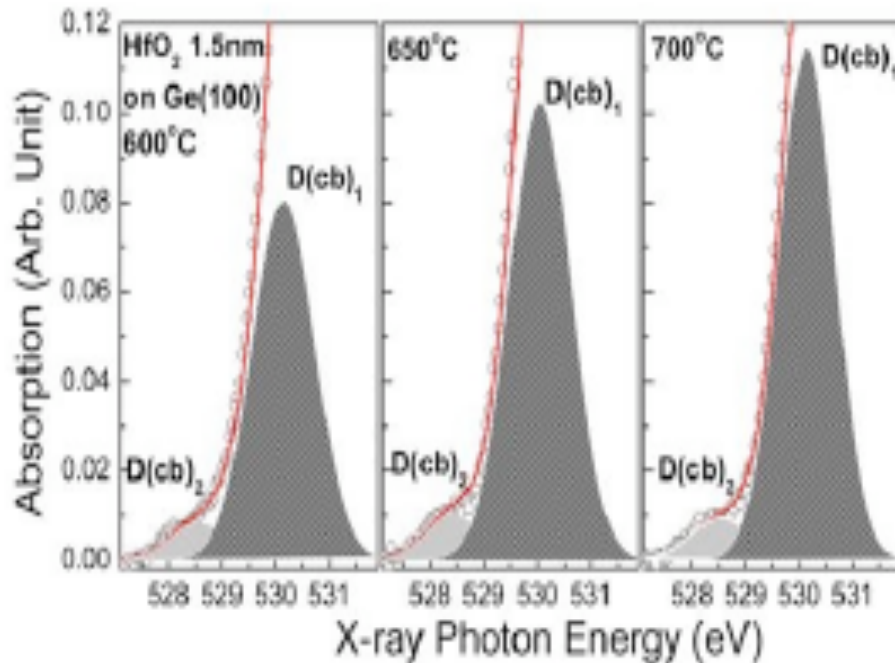
Ge-O sublimates ~400-500°C
Ge-N decomposes ~600-650°C

solid phase epi-regrowth
of Ge on Ge substrate
nano-grain “reorganization”
textured films with “c-axis”
aligned relative to dimer rows
of Ge substrate

we will focus on two aspects of this process

i) verification of process variations that result in HfO₂ and HfSiON in “direct contact” with Ge(0001); i.e., no detectable Ge dielectrics forming an interfacial transition region

ii) relationship between processing and defects extracted from MOSCAP test devices

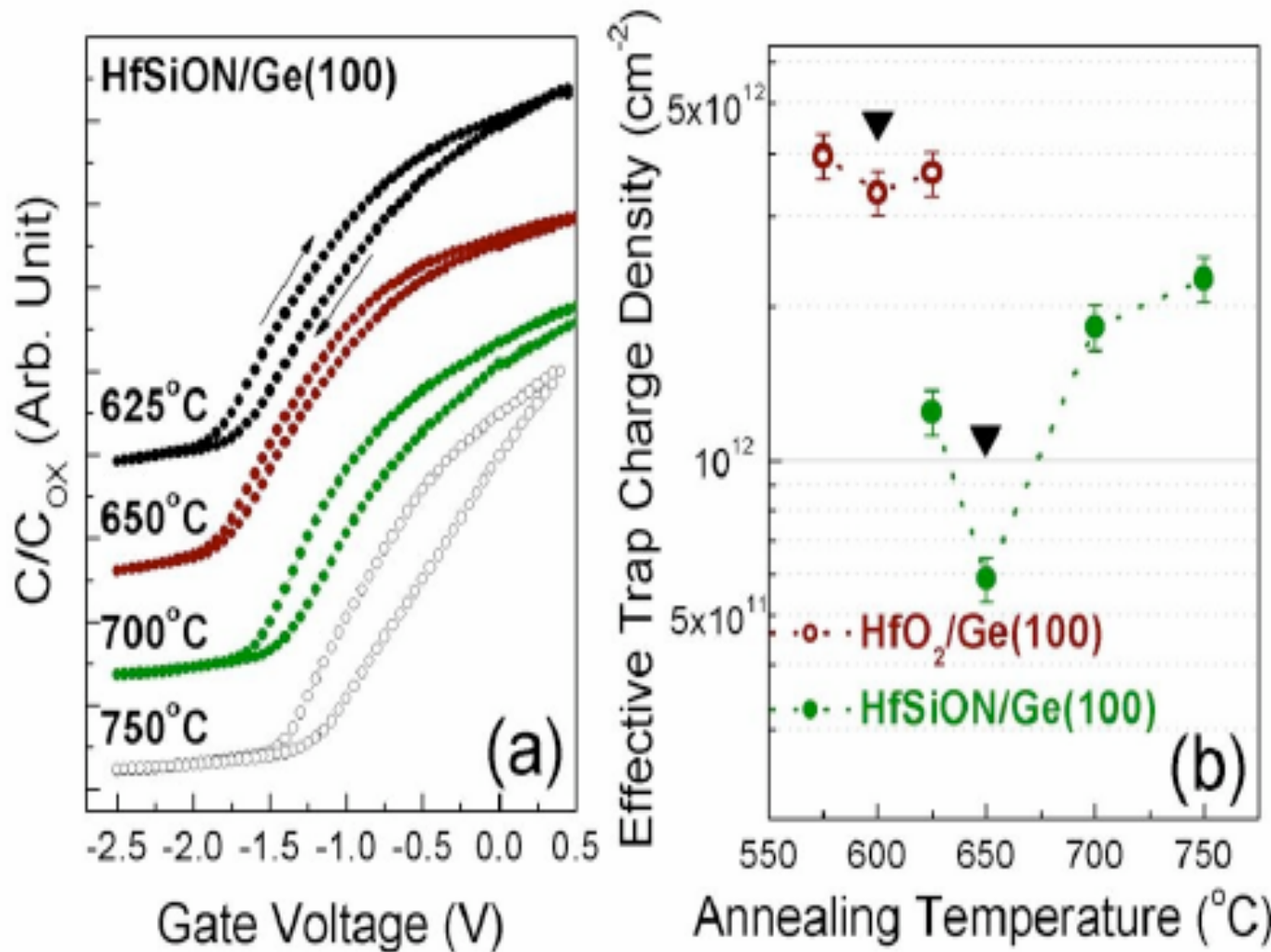


(a) Thermal evolution of the O K_1 spectra for both HfO_2 and HfSiON over a narrow energy region below the doubly-degenerate Hf 5d conduction band edge states as a function of PDA temperature.

two deconvoluted peaks, labeled $\text{D}(\text{cb})_1$ and $\text{D}(\text{cb})_2$, indicate the defect states determined by Gaussian fits of the XAS O K_1 edge spectra.

lower density of defects by spectroscopy for HfSiON than HfO_2 - confirmed by CV

Capacitance-voltage (CV) for *n*-MOSCAPS 5nm HfSiON on *n*-type Ge (100): normalized by C/C_{ox} , as a function of PDA temperature-arrows indicate clockwise hysteresis loop.



lowest trap density same as Saraswat's (Stanford Univ) best

HfSiON on Si - X-ray stress similar to SiO₂ no negative fixed charge as in HfO₂

defects lower than in HfO₂ agree with XAS

conclusions with respect to

CMOS

(i) HfO₂ and Hf Si oxynitride on Si: there is a direct correlation between intrinsic bonding defects observed in spectroscopic studies with those obtained in electrical measurements on MOSCAPs

(i) HfO₂ and Hf Si oxynitride on Ge: due to limitations on processing temperatures, generally lower than 800C, often more restrictive to 650C, CMOS applications are generally more limited

currently been pursued

rad testing

CMOS - first important result under DTRA radiation induced defects can be observed spectroscopically in XAS, spectroscopic ellipsometry and cathodoluminescence spectroscopy

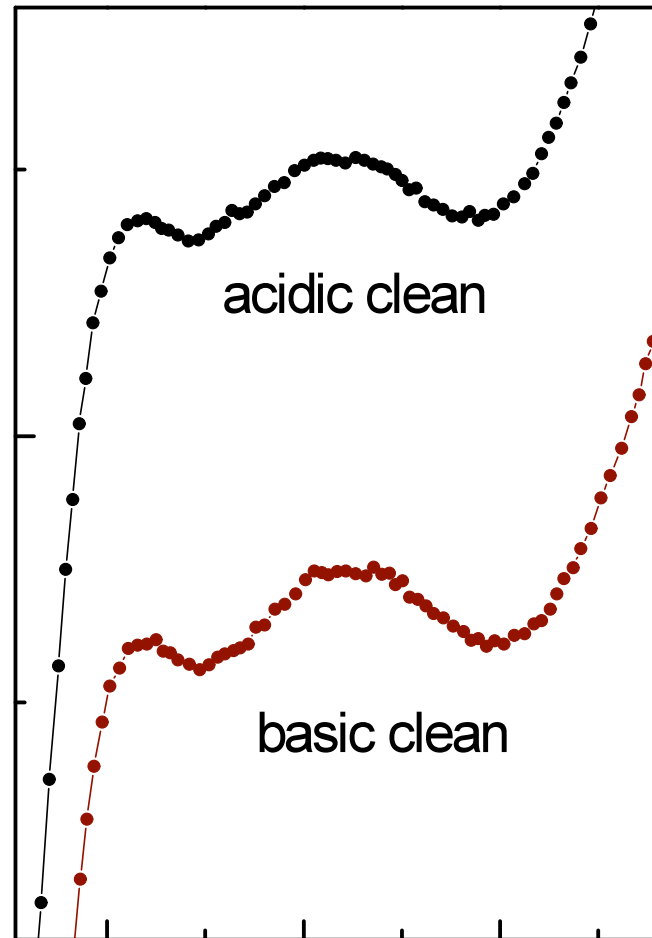
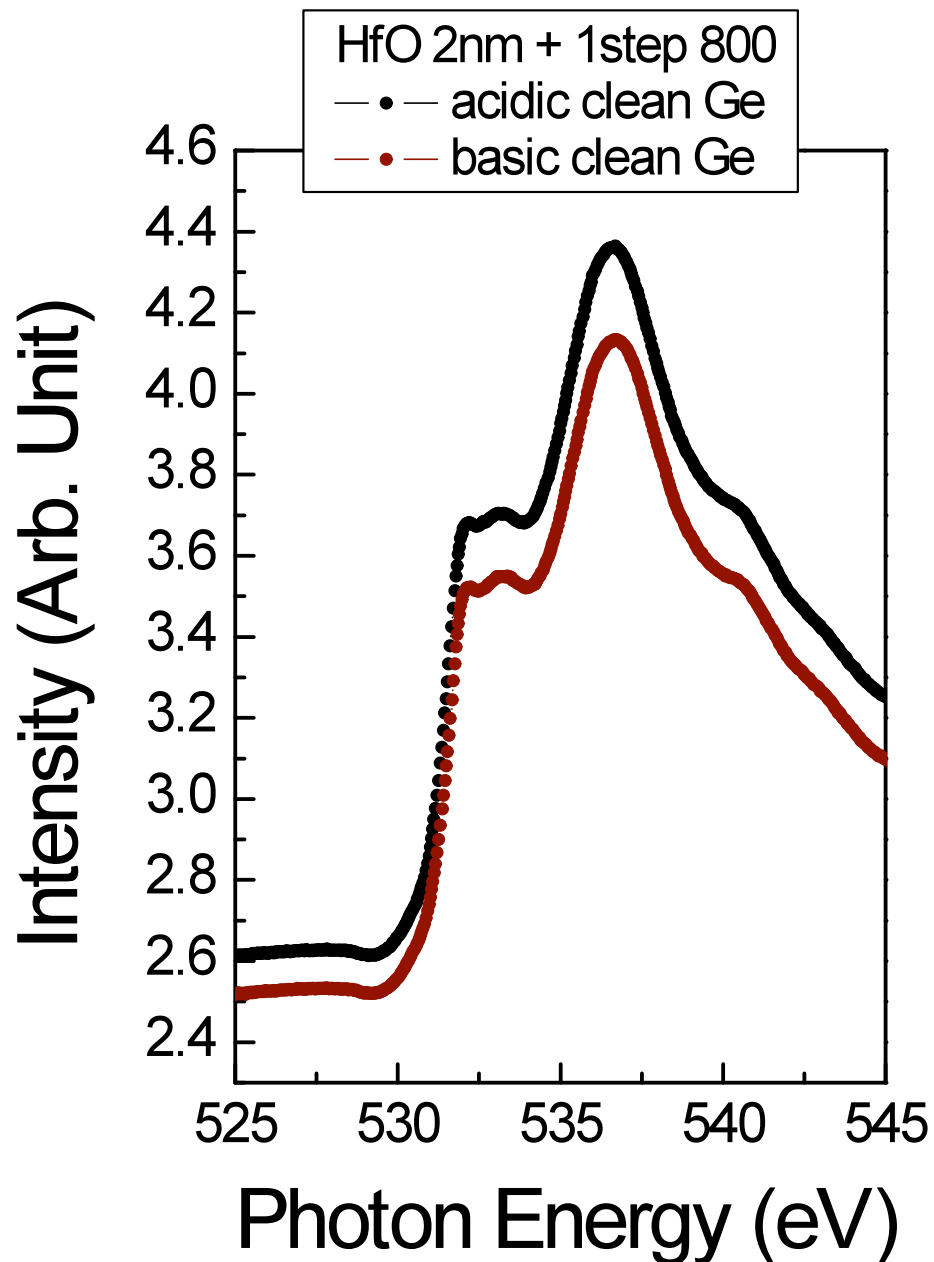
plans for next year

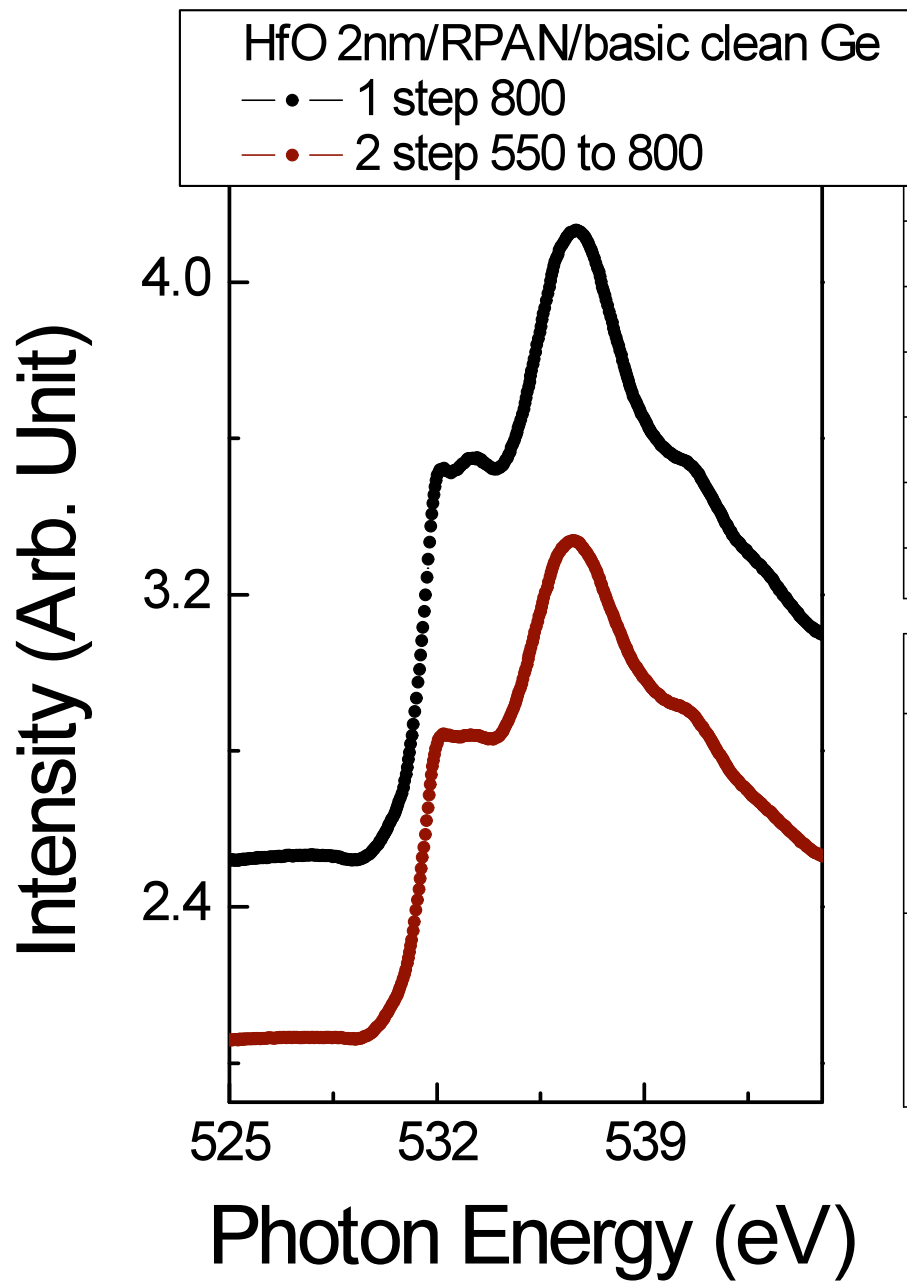
- (i) prepare MOSCAPs on Ge(100) and Si/SiON/ substrates for rad testing using optimized Hf Si Oxynitride composition
(HfO₂)_{0.3}(SiO₂)_{0.3}(Si₃N₄)_{0.4}**
- (ii) fabricate stacked HfO₂/HfSiON on Ge(100) MOSCAPs using the HfSiON as a thin interfacial transition and compare electricals with HfO₂ and HfSiON on Ge(100)
determine defect concentrations, and electrical stress induced defects**
- (iii) Based on results from (ii) prepare another set of MOSCAPs for rad testing and compare defects inducted by radiation and electrical stress**
- (iv) perform spectroscopic studies to compare electrically active defects with defects observed spectroscopically**

Gerry Thanks You

1 step monoclinic HfO₂

textured epi-growth -
independent of acidic or basic



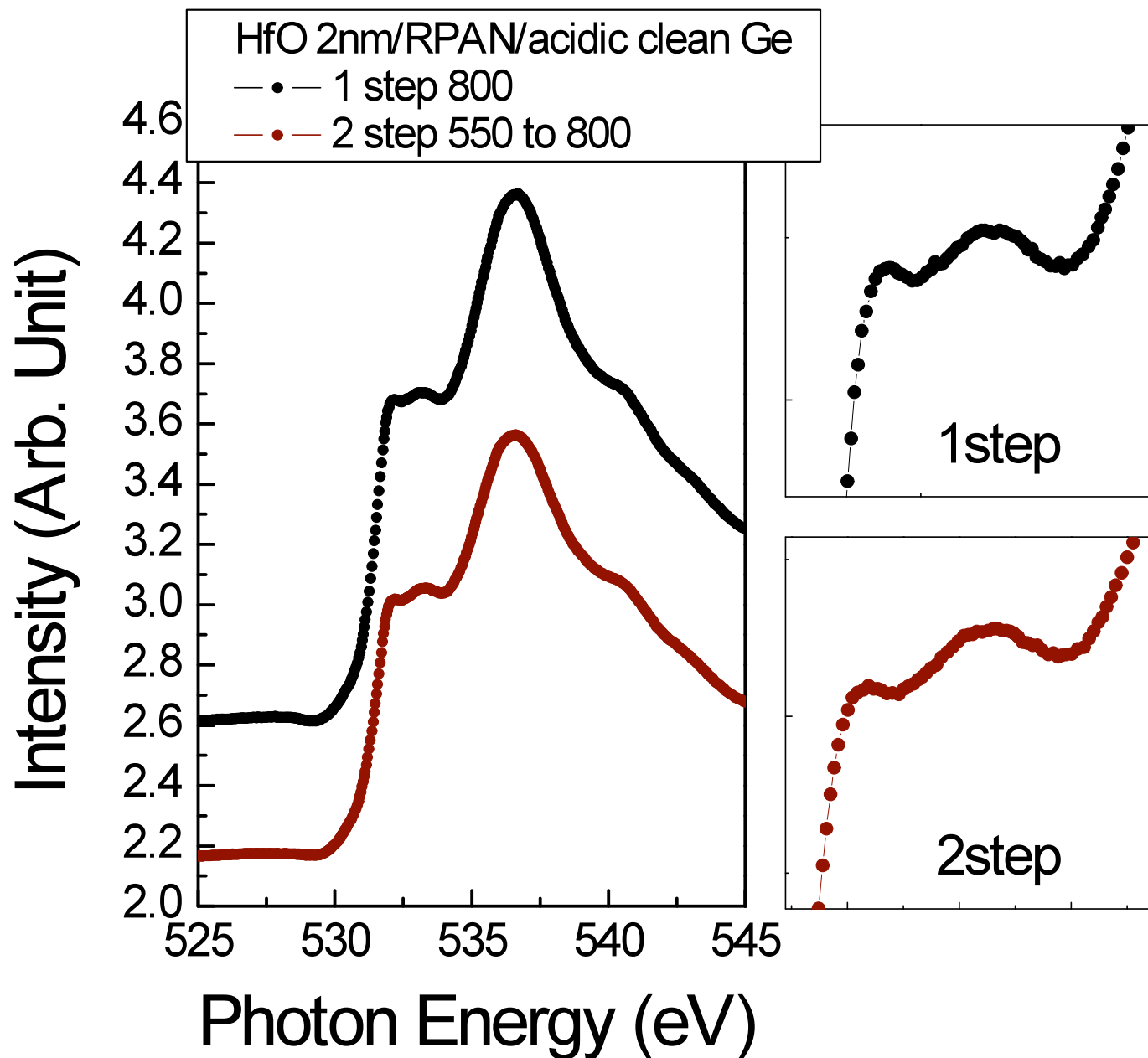


**1 step
monoclinic
HfO₂**

**textured epi-
growth - aligned
to dimer rows of
Ge(100)**

**2 step mixed
monoclinic-
tetrahedral
HfO₂ grains**

**Ge-Hf oxide
dielectric at
interface after
anneal**

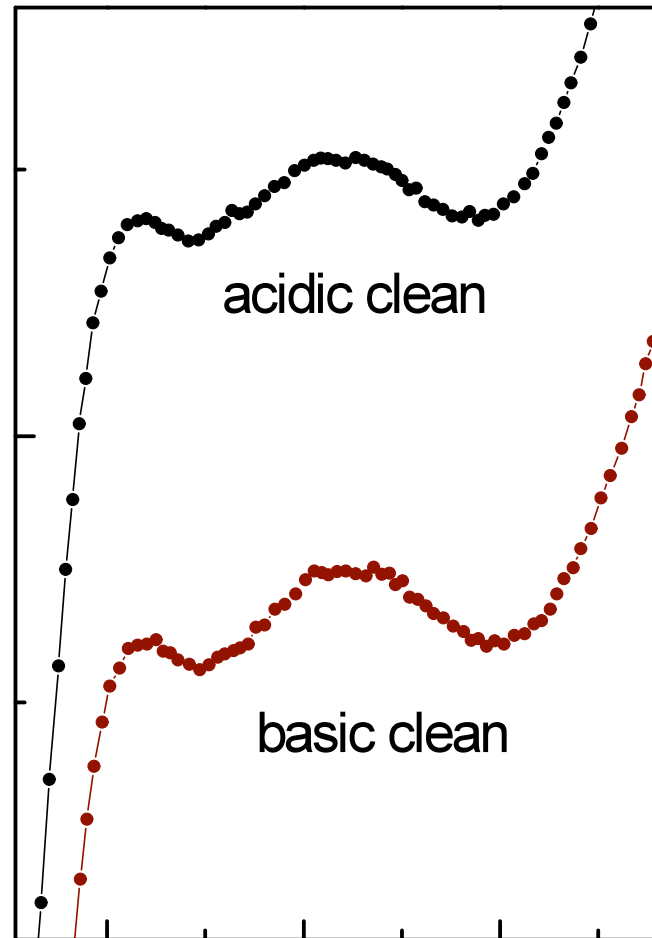
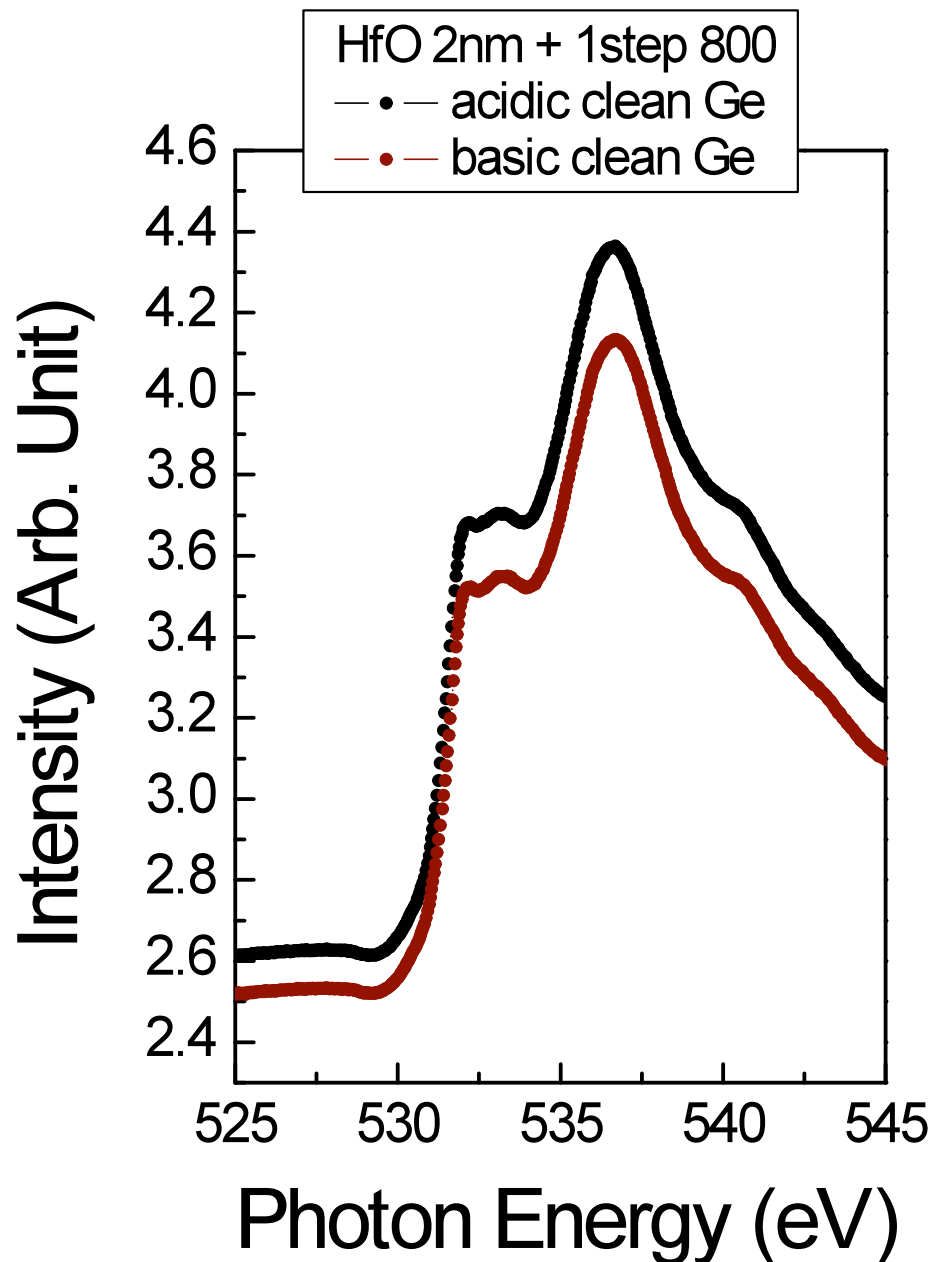


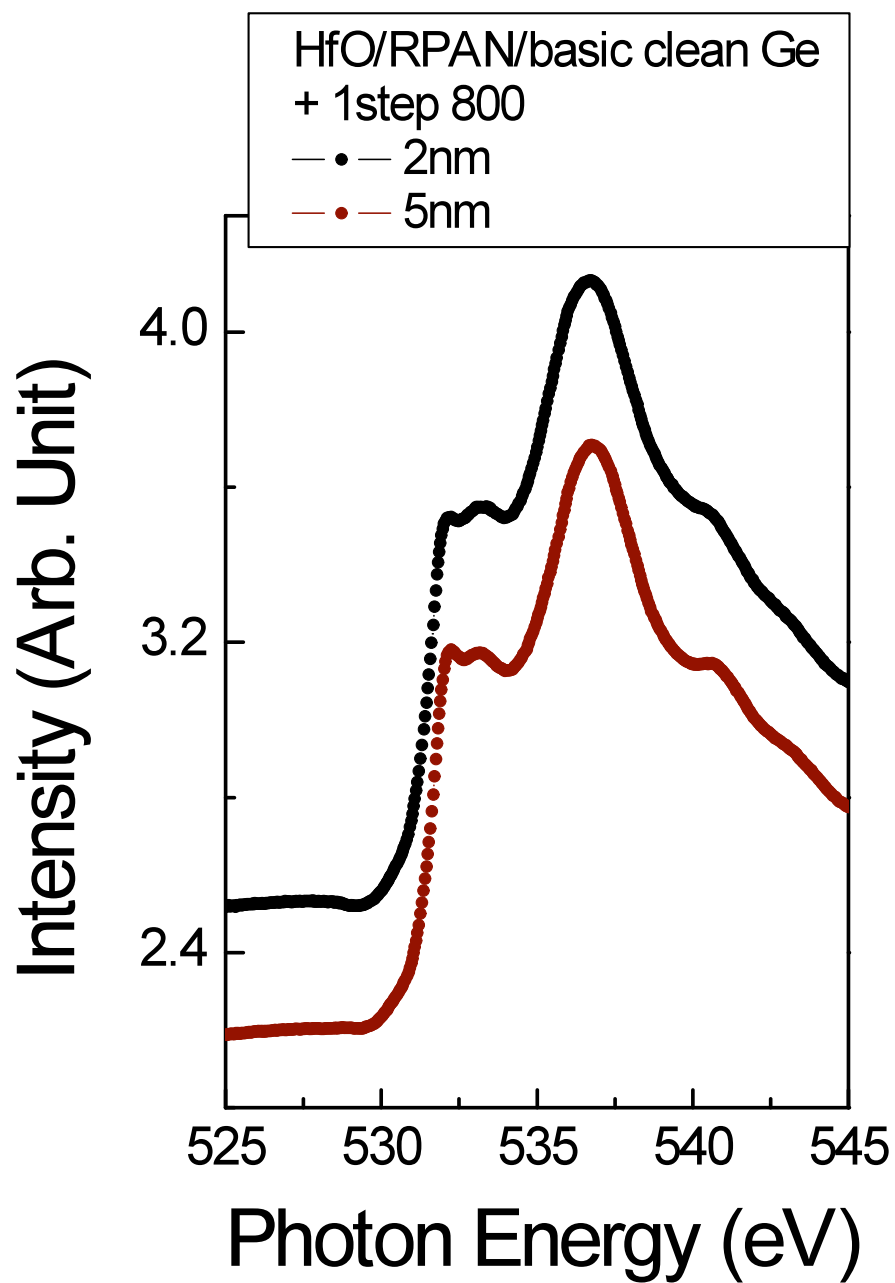
**1 step and two
step with acidic
clean give
monoclinic
HfO₂**

**textured epi-
growth - aligned
to dimer rows of
Ge(100)**

1 step monoclinic HfO₂

textured epi-growth -
independent of acidic or basic



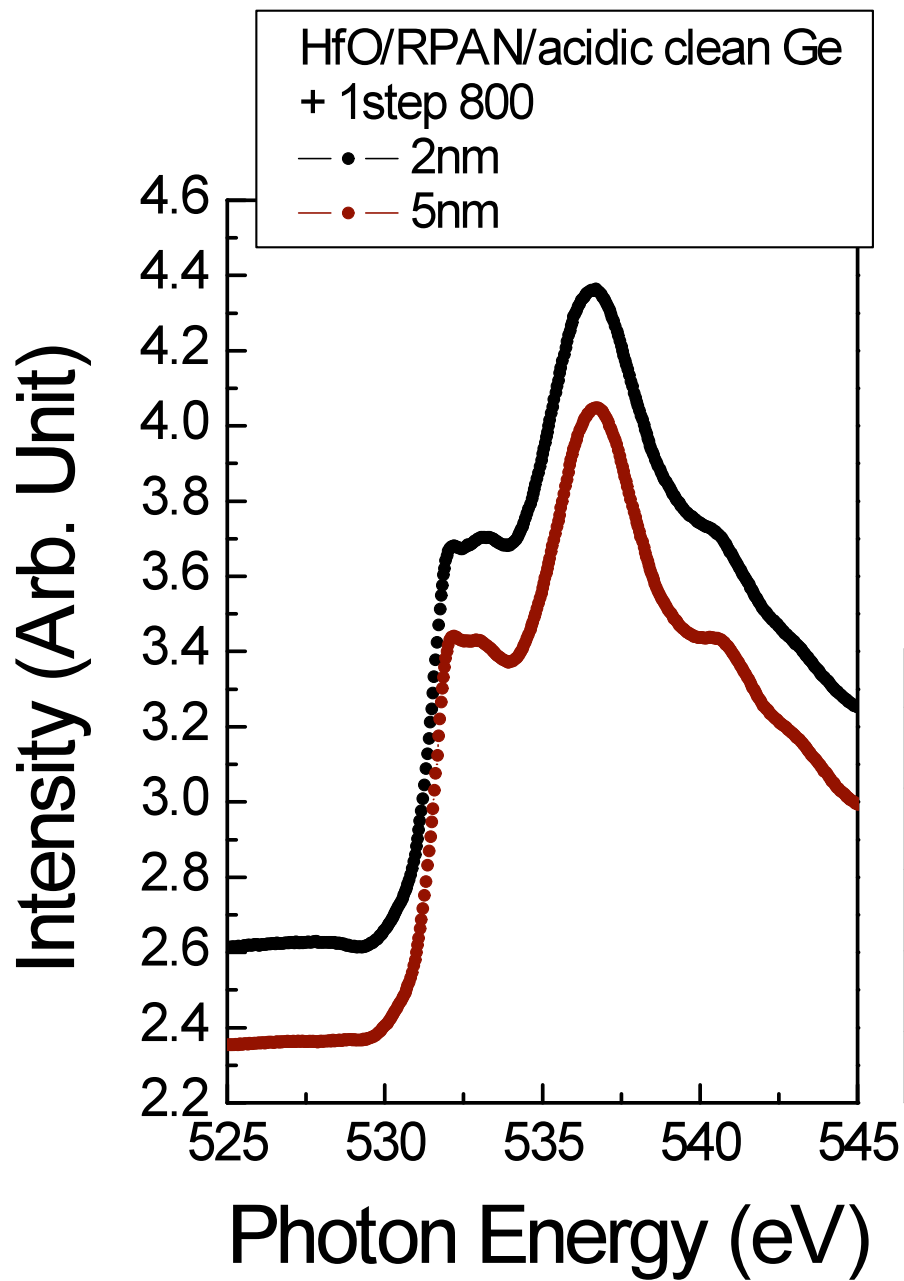


**2nm 1 step
monoclinic HfO₂**

**textured epi-growth
- aligned to dimer
rows of Ge(100)**

**5nm 1 step
t- HfO₂ on m-HfO₂**

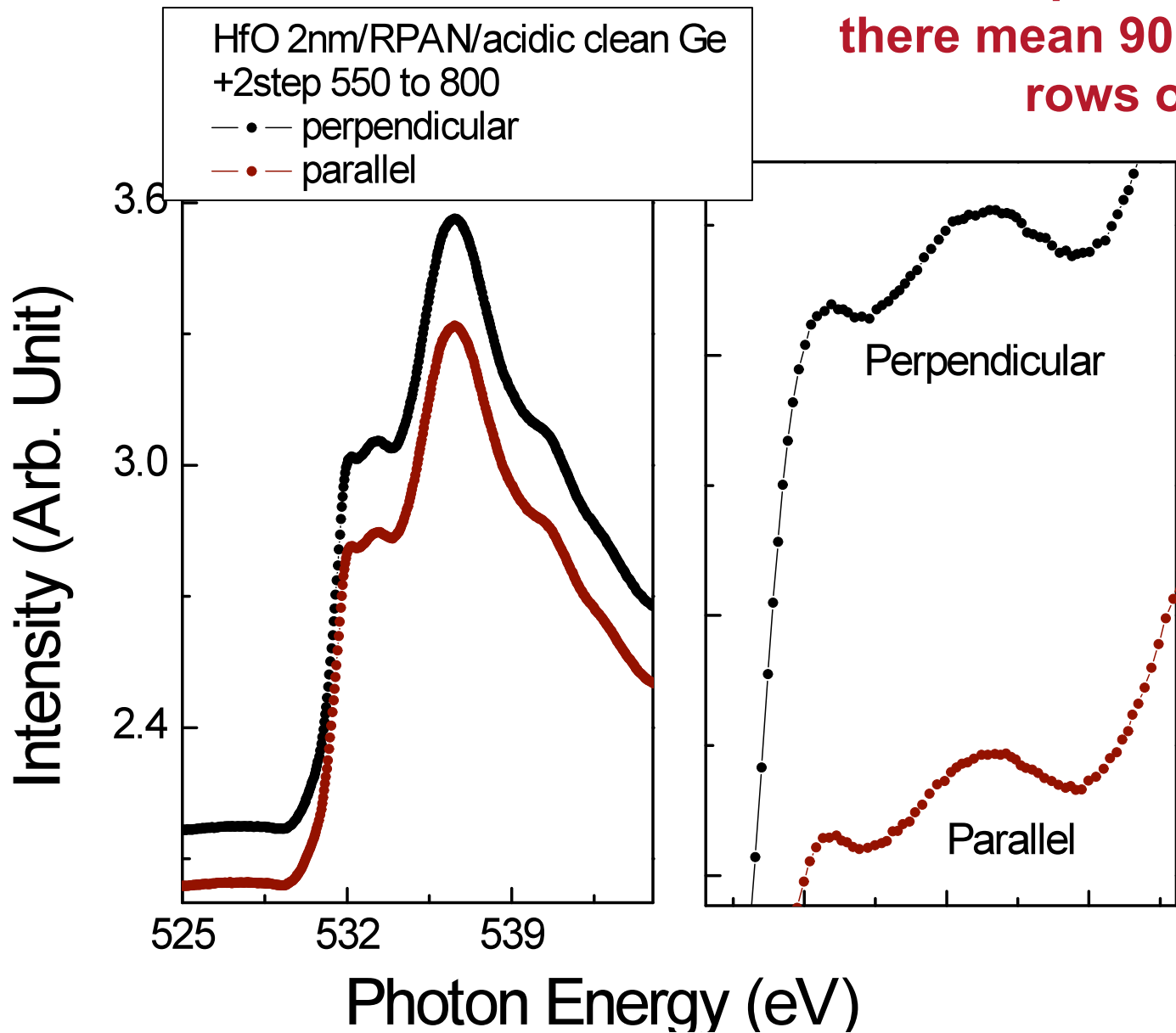
**as film grows
mixed m- and t-
grains are formed -
growth loses
contact with Ge
surface geometry**



**2nm 1 step
monoclinic HfO₂**

**5nm 1 step
t- HfO₂ on m-HfO₂**

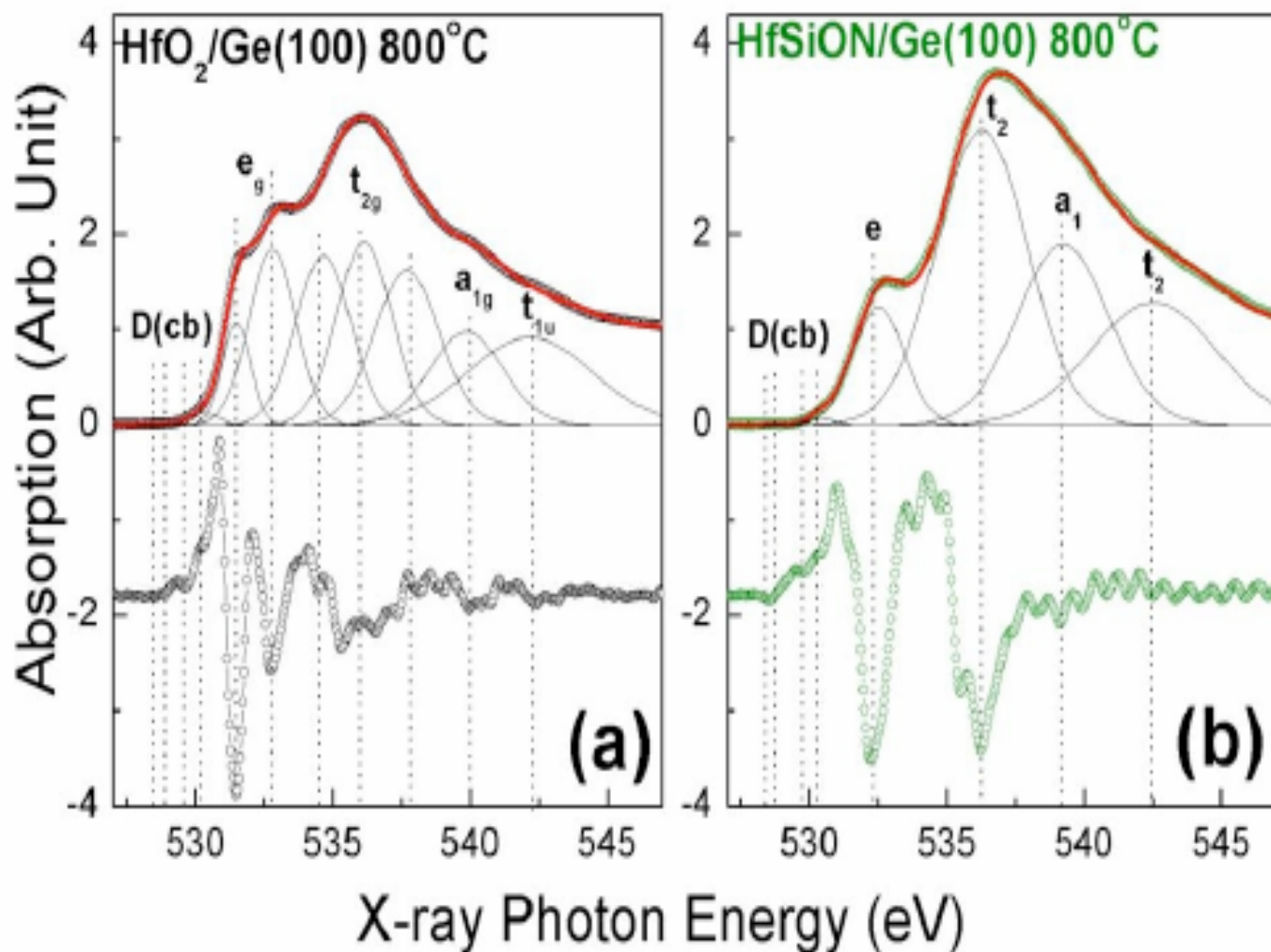
**mixed grain growth
in thicker films
independent of
surface clean**

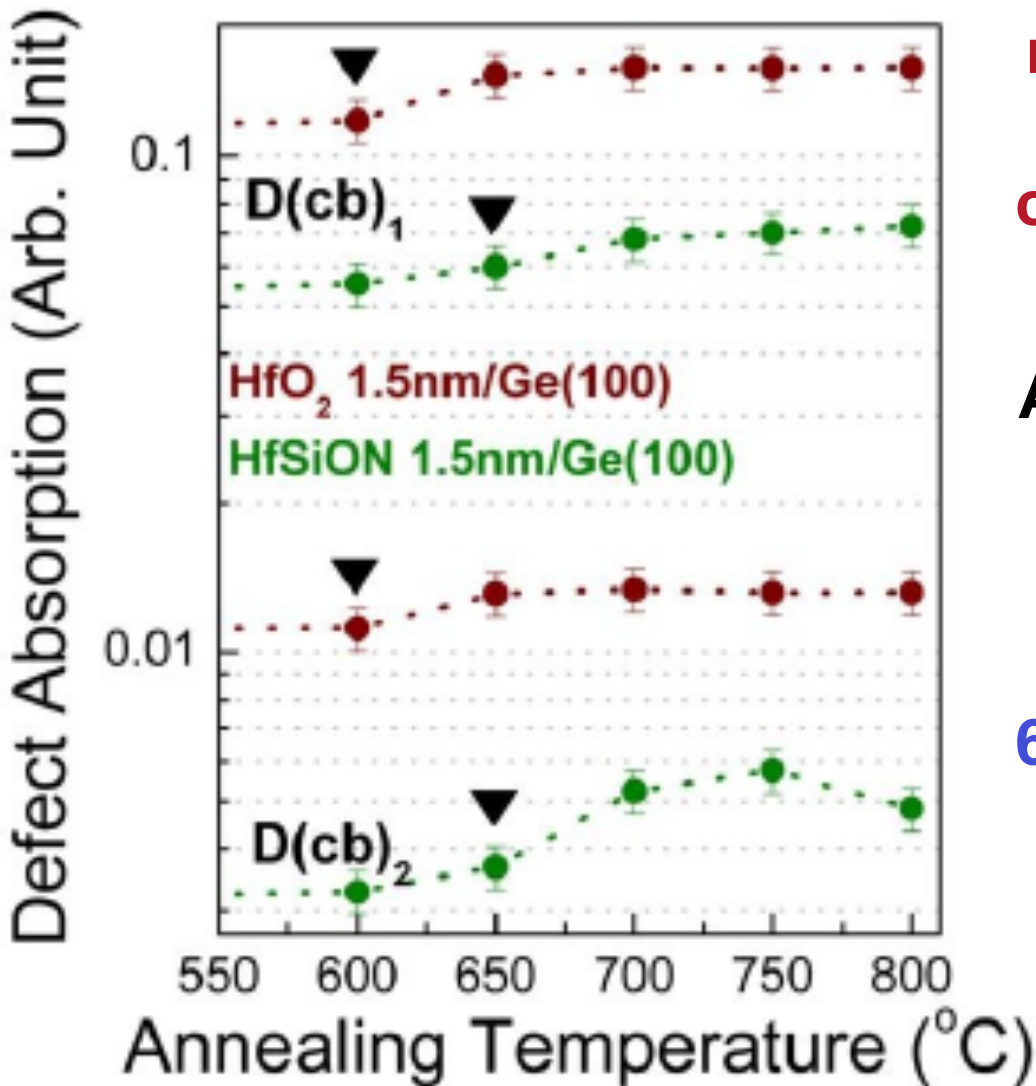


perpendicular and parallel are with respect to wafer flat, and there mean 90 rotation of dimer rows of Ge(001)

since both spectra are the same - axis of unit cell for alignment makes an angle of 45 degrees with respect to dimer rows

XAS O K_1 edge: top (a) HfO_2 on Ge-(b) HfSiON on Ge BottomL 2nd derivative O K_1 edge. Gaussian fits: energies 2nd derivatives give, crystal field (C-F) splittings Jahn-Teller (J-T) d -state degeneracy removal: molecular orbital labels: local bonding symmetry-group theory. defect states below the conduction band edge ~ 530 eV from 2nd derivative spectra - Gaussian fits next slide





thermal evolution of relative absorption for two distinct defect states, calculated by Gaussian fits in previous slide

Arrows indicate the onsets of distinct changes in these trends for each defect state.

600-650 $^{\circ}\text{C}$ release of N from GeN as observed in N K edge spectra

Differences Between Charge Trapping States in Irradiated Nano-crystalline HfO₂ and Non-crystalline Hf Silicates

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USA

Supported in part by ONR, SRC, and an AFOSR MURI

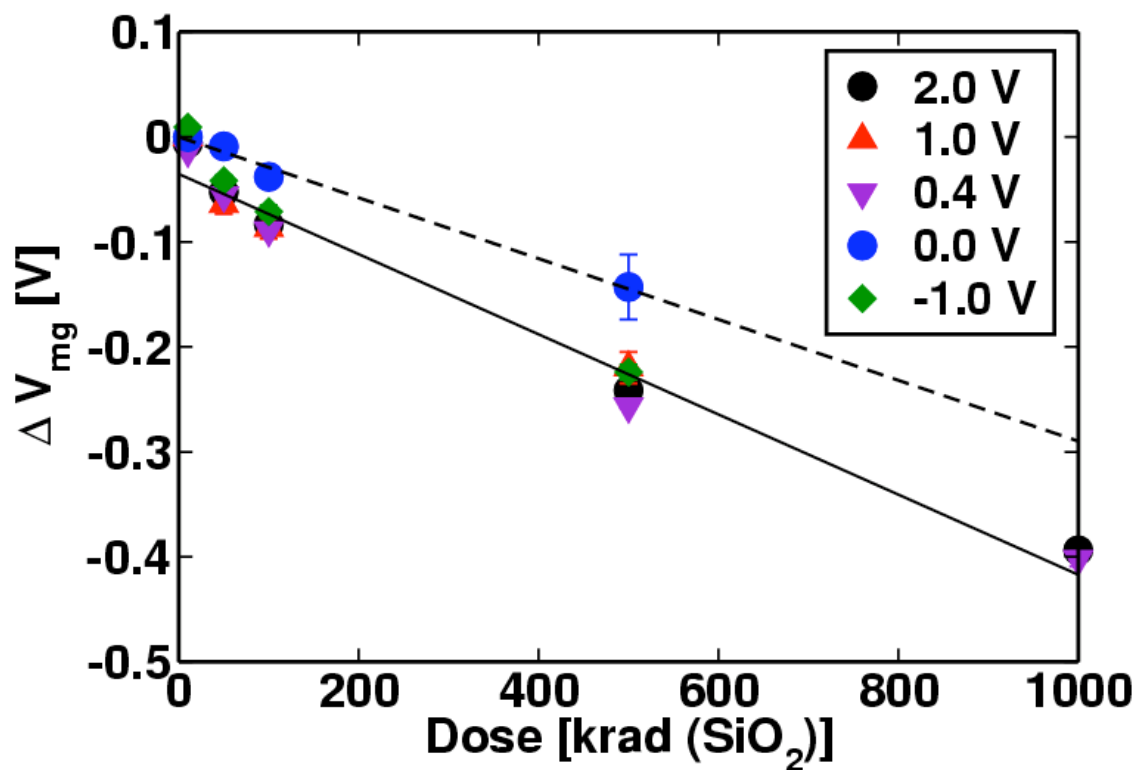
³Sandia is a multiprogram laboratory operated by Sandia Corporation, a Lockheed Martin Company, for the United States Department of Energy's National Nuclear Security Administration under Contract DE-AC04-94AL85000.

Outline

- **Experimental results on hole and electron trapping**
 - **Hf silicate (amorphous)**
 - **HfO₂ (nano-crystalline)**
- **New spectroscopic evidence**
 - **Hf and other group IV B oxides**
 - **Soft x-ray photoelectron spectroscopy (SXPS)**
- **Insight into differences into electron and hole trapping**
- **New promise for advanced high-κ dielectrics**

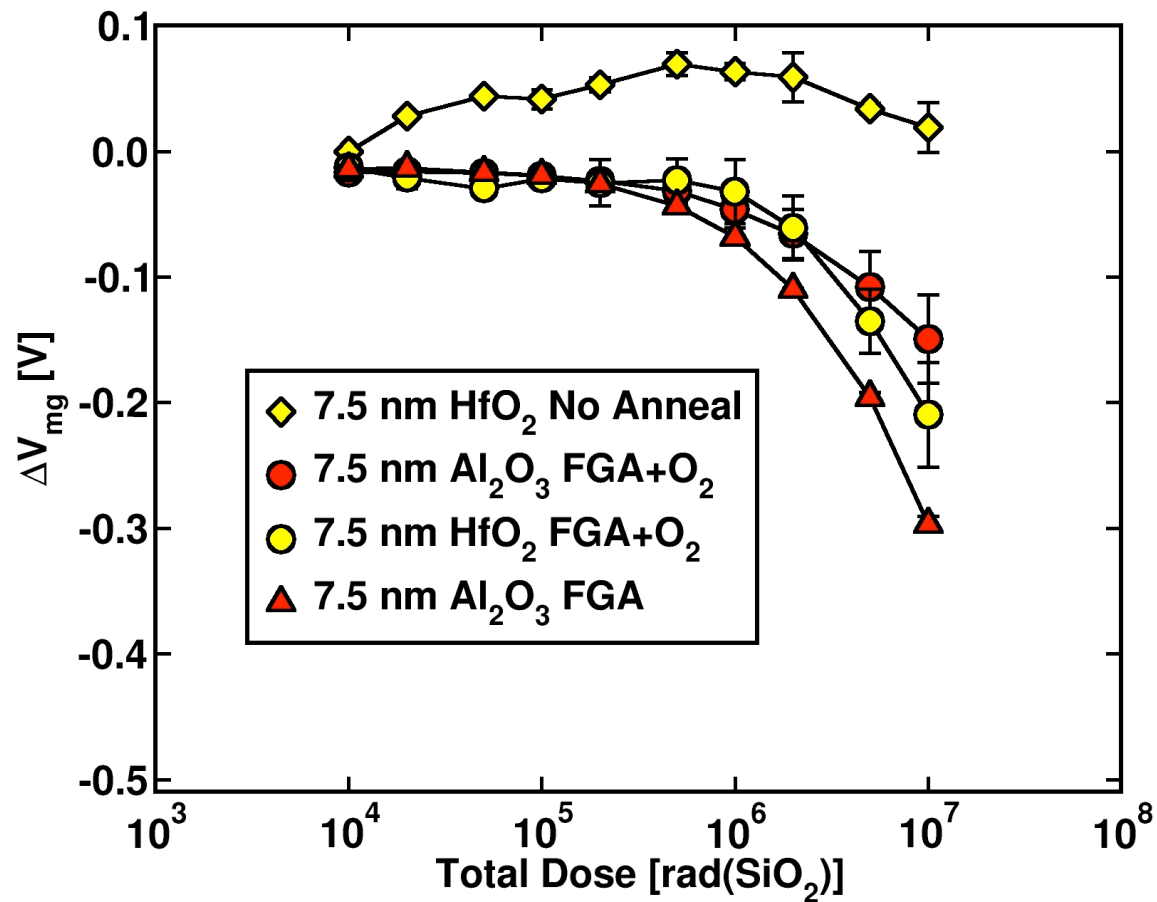
Radiation response of Hf Silicate capacitors

Mostly hole trapping in oxide



J. A. Felix, D. M. Fleetwood, R. D. Schrimpf, J. G. Hong, G. Lucovsky, J. R. Schwank, and M. R. Shaneyfelt, *IEEE Trans. Nucl. Sci.*, vol. 49, pp. 3191-3196, 2002.

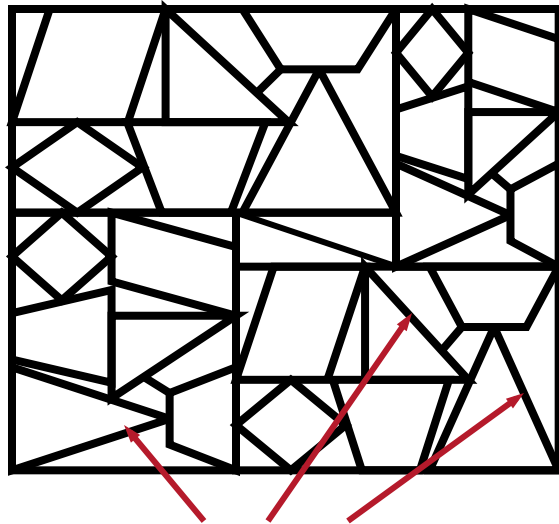
Radiation response of HfO₂ capacitors Both hole and electron trapping



J. A. Felix, J. R. Schwank, D. M. Fleetwood, M. R. Shaneyfelt, and E. P. Gusev, *Microelectron. Reliab.*, vol. 44, pp. 563-575, 2004.

Micro-structure of nano-crystalline HfO₂ and non-crystalline Hf silicates and nitrided Hf silicates

nano-crystalline HfO₂



grain boundaries

crystallite size: 2 to 100 nm
grain boundaries - intrinsic bonding defects
local defects at band edges
trap both holes and electrons

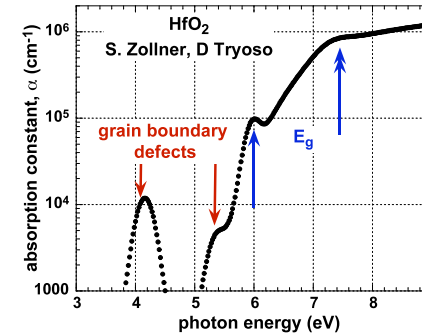
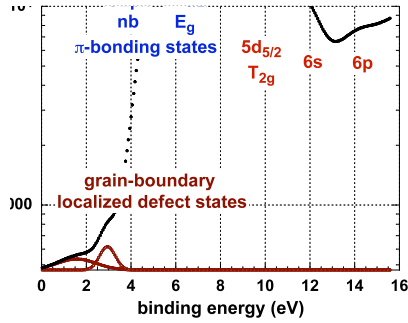
amorphous Hf silicate



isotropic and homogeneous
no discernable granularity
band tail states due to random local atomic structure
localized defect states
generally deep in gap
specific bonding defects
broken bonds - chemical impurities

Spectroscopic evidence for band edge defects in HfO₂

Soft x-ray photoelectron spectroscopy (SXPS)



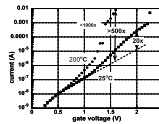
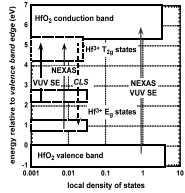
partially-filled defect states at
valence band edge

hole or electron traps

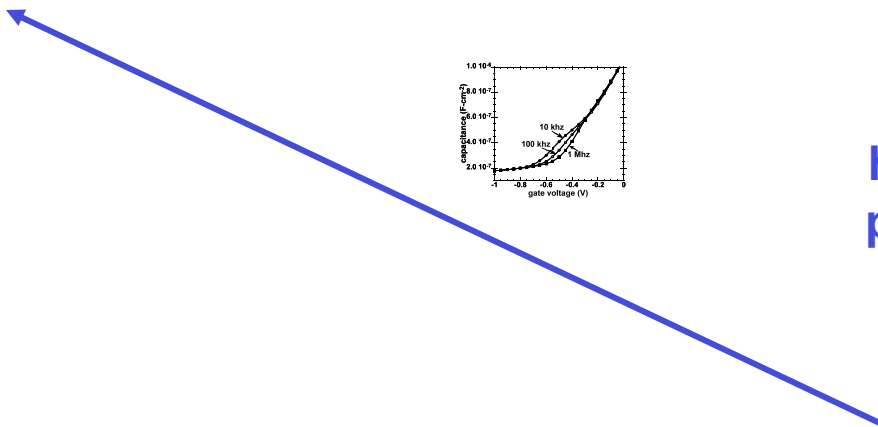
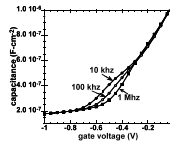
empty defect states at
conduction band edge

electron traps

Intrinsic band edge defect states function as electron and hole traps in a radiation or high-field stress environment



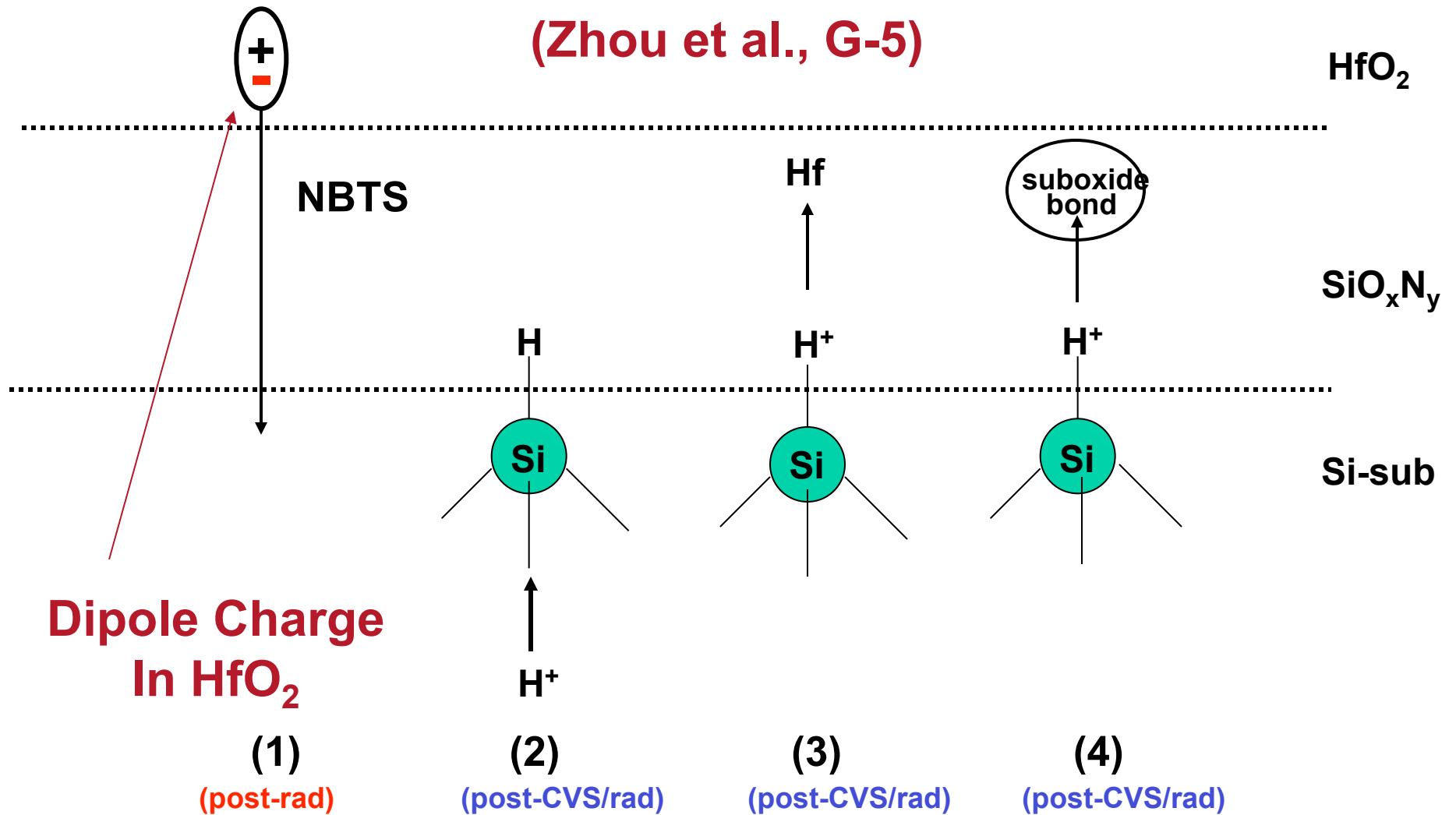
trap-assisted transport of electrons through empty states



hole trapping into partially-occupied states

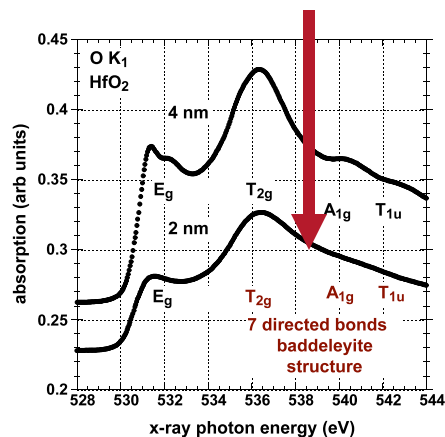
Mechanisms during post-rad/ CVS NBTS on HfO₂ capacitors

(Zhou et al., G-5)



Ultra-thin HfO_2 is a valid engineering solution for advanced devices

band edge states change dramatically
in ultra-thin HfO_2 thin films
for ≤ 2 nm physical thickness
no Jahn-Teller effects and no
band edge states in SXPS
Sematech, and ST Micro
report dramatic drops in
defects as well

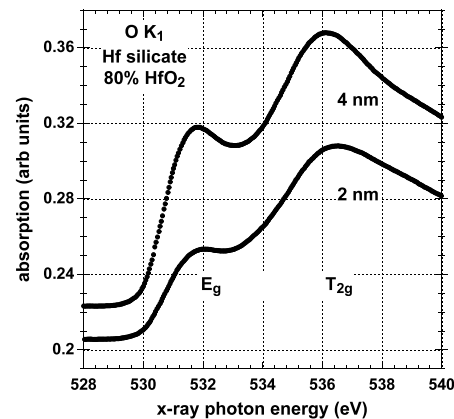


**Ultra-thin phase separated 80% HfO₂: almost promising
(nano-crystalline HfO₂ with nano-scale SiO₂ inclusions)**

**band edge J-T term-splittings
suppressed by small sizes
(<2 nm) in nano-crystalline
grains due to SiO₂ inclusions**

**2 nm thick spectra same as
2 nm thick HfO₂ on last slide**

**4 nm thick sample - grains
are anisotropic in growth
direction -- therefore
narrowing of E_g feature**



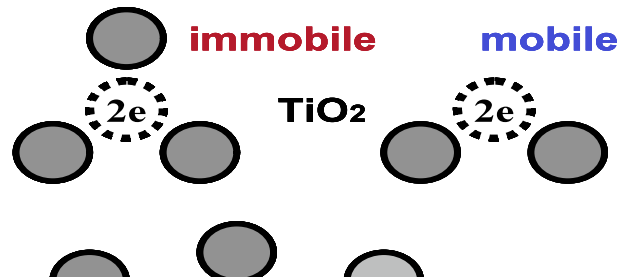
We await results from electrical tests in full devices!

Summary and Conclusions

- **Hf silicate and HfO₂ gate dielectrics show differences in electron trapping**
- **Spectroscopic evidence shows enhanced electron trapping in HfO₂ caused by nano-crystallinity**
- **Significant issue – especially for combined radiation/reliability response**
- **New Hf, Zr oxynitrides and thin HfO₂ films (with/without SiO₂ inclusions) may be promising for highly scaled, future technologies**
 - **Spectroscopic studies show absence of electron traps**
 - **Await the results of fully processed devices**

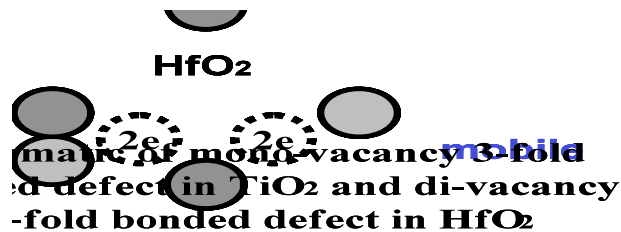
schematic representation of immobile and mobile vacancy defect bonding

removal to one neutral O-atom - 2 electrons/O in vacancy site



TiO₂ local bonding in distorted rutile geometry -- 6 fold coordinated Ti and 3 fold coordinated O

immobile 3 Ti; mobile 2Ti

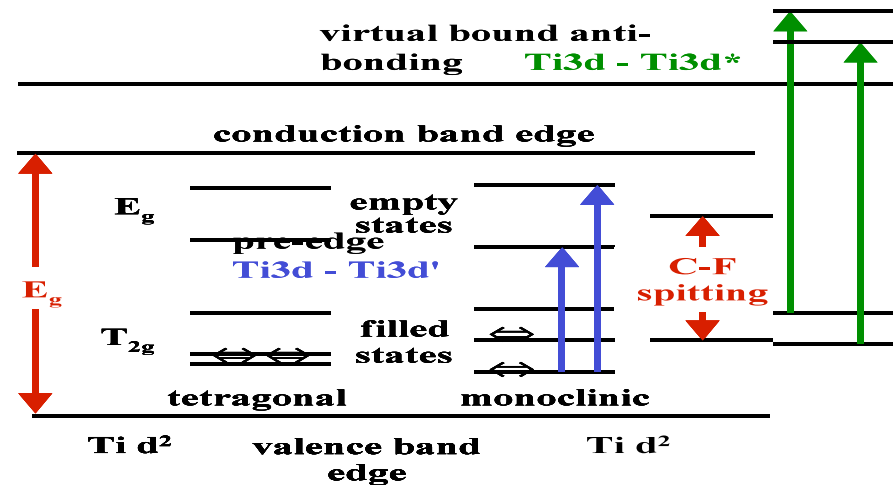
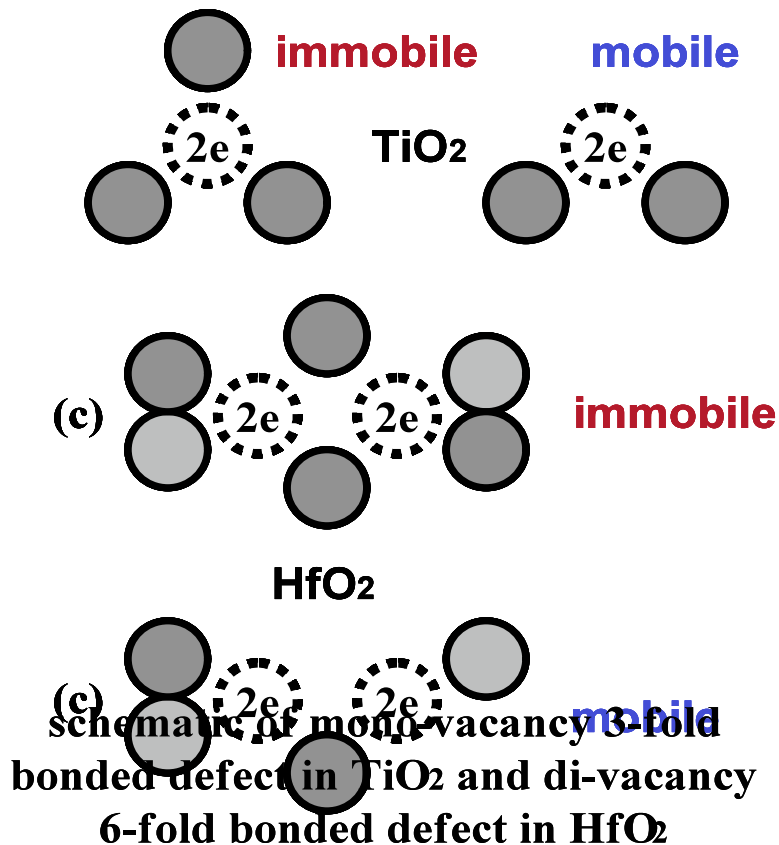


HfO₂ local bonding in distorted CaF₂ geometry -- 8 fold coordinated Hf and 4 fold coordinated O

immobile 6 Hf; mobile < 6Hf, e.g., 4 or 5 Hf

mono-vacancy (or simply vacancy) defects

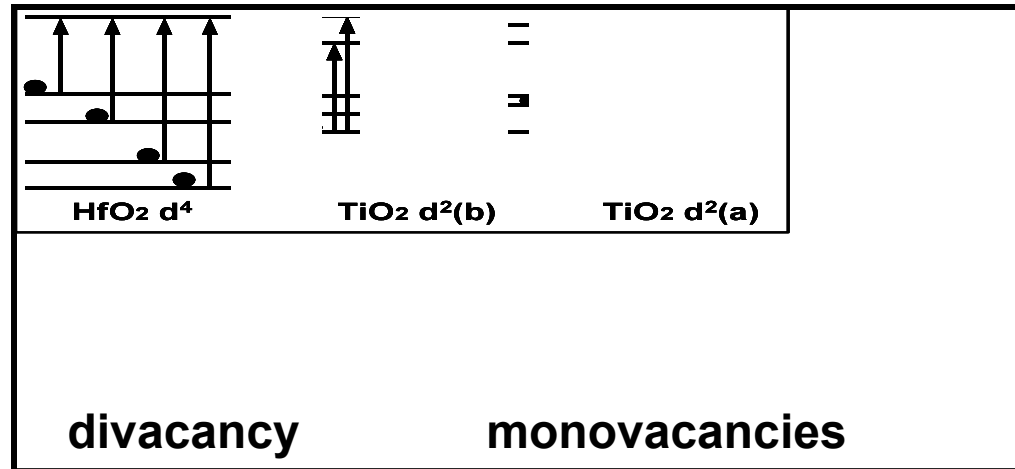
2 states at band edge in SXPS TiO₂ - consistent with d² description
 XAS pre-edge & virtual bound state regimes - additional confirmation



mono-vacancies are d² -- di-vacancies are d⁴

difference between immobile and mobile - spectral widths proportional to crystal field - number of atoms bordering on defect

summary of d-d' transitions* for HfO₂ d⁴ divacancies and TiO₂ d² mono-vacancies



HfO₂ immobile and TiO₂ (a) = mobile and (b) = immobile

*common mistake -- assumption that these transitions originate at top of VB places the final states too deep in the gap (SEMATECH group at WoDiM 2008, and T-J King's group in JVST paper)
 significant issue for interpretation of trap level for trap assisted tunneling and/or PF transport